# Parasite glycoconjugates. Part 11. ${ }^{1}$ Preparation of phosphodisaccharide synthetic probes, substrate analogues for the elongating $\alpha$-d-mannopyranosylphosphate transferase in the Leishmania 

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A set of phosphodisaccharides, substrate analogues, which will be used to study acceptor-substrate specificity of the Leishmania biosynthetic enzymes, are synthesized using the Koenigs-Knorr and trichloroacetimidate methods for the glycosylation reactions, $S_{\mathrm{N}} 2$ nucleophilic displacement of a triflic ester for epimerization, and the glycosyl hydrogenphosphonate method for phosphorylation.

## Introduction

Throughout the tropics and subtropics Leishmania parasites cause a variety of diseases ranging from self-limiting skin lesions to the often fatal visceral leishmaniasis. The surface lipophosphoglycan (LPG) produced by the infectious promastigote stage of all species of the Leishmania contains a polymeric section consisting of $(1 \rightarrow 6)$-linked $\beta$-d-galactosyl$(1 \rightarrow 4)-\alpha$-D-mannosyl phosphate repeating units. The importance of the LPG for parasite infectivity and survival ${ }^{2}$ makes the enzymes responsible for the biosynthesis of this glycoconjugate of great interest. Phospho-oligosaccharide fragments of the LPG of L. donovani, L. mexicana and L. major were synthesized ${ }^{3-6}$ in our laboratory and tested as acceptor substrates (in vitro) for the Leishmania $\alpha$-d-mannopyranosylphosphate transferase (MPT) responsible for the transfer of $\alpha$-D-Man $p$ phosphate from GDP-Man to the growing phosphoglycan chain. It was shown ${ }^{7}$ that the phosphodisaccharide $\mathbf{1}^{4,8}$ (representing one repeating unit of the phosphoglycan) is the minimal structure exhibiting acceptor substrate activity for the MPT.
In Part $9^{8}$ of this series, we disclosed our interest in the design and synthesis of various structural analogues of compound 1 to test the fine acceptor substrate specificity of the MPT and to gain more information about enzyme-substrate recognition. Thus, phosphodisaccharides 2-5, which are epimers of the substrate $\mathbf{1}$ at $\mathrm{C}-1^{\prime}, \mathrm{C}-2^{\prime}, \mathrm{C}-3^{\prime}$ or $\mathrm{C}-4^{\prime}$, respectively, have been synthesized.

We now report the chemical synthesis of the disaccharide phosphates 6-10. Compounds $\mathbf{6}$ and $\mathbf{7}$ are epimers of the sub-
strate $\mathbf{1}$ at C-2 and C-3, respectively, of the D-mannopyranose moiety. Compounds $\mathbf{8}$ and $\mathbf{9}$ are substrate analogues deoxygenated at positions C-6 and C-6', respectively. In this context, the preparation of the analogue $\mathbf{1 0}$, which is an epimer of compound 9 at $\mathrm{C}-1^{\prime}$ and could be (as well as the analogue 9 itself) a potential inhibitor of the enzyme, is also described. The information obtained from testing the acceptor activity of the substrate analogues 2-10 will be used to predict which sugar hydroxy groups of compound $\mathbf{1}$ are involved in enzymesubstrate recognition events and to design potential enzyme inhibitors.

## Results and discussion

The synthetic schemes for the preparation of the phosphodisaccharides 6-10 consist of a few general steps (Scheme 1): 1) synthesis of fully protected disaccharide derivatives $\mathbf{A} ; 2$ ) anomeric de-O-protection ( $\longrightarrow \mathbf{B}$ ); 3) H-phosphonylation at position O-1 ( $\longrightarrow \mathbf{C}$ ); 4) coupling of the H-phosphonates $\mathbf{C}$ with dec-9-en-1-ol (using the glycosyl H-phosphonate method) ${ }^{9}$ to furnish the protected glycosyl phosphodiesters $\mathbf{D}$; 5) total de- $O$-protection.

The octa- $O$-acetyl- $\alpha, \beta$-lactose 11 ( $\alpha: \beta=7: 1$; which is a precursor of the phosphodisaccharide 6; Scheme 2) was prepared by conventional acetylation of $\alpha$-lactose and then converted to the hemiacetal $12(83 \% ; \alpha: \beta=4: 1)$ by anomeric de- $O$-acylation ${ }^{3-6,8-10}$ with dimethylamine in $\mathrm{CH}_{3} \mathrm{CN}-$ THF. H-Phosphonylation ${ }^{3-6,8-10}$ of compound 12 with triimidazolylphosphine (prepared in situ from $\mathrm{PCl}_{3}$, imidazole and $\mathrm{Et}_{3} \mathrm{~N}$ ) followed by mild hydrolysis produced a mixture of $\alpha$ - and $\beta$-linked


$\beta$-D-Galp-( $1 \rightarrow 4$ )- $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}$-OR

$\alpha$-D-Galp-(1 $\rightarrow 4$ )- $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}$-OR

$\beta$-D-Talp- $(1 \rightarrow 4)$ - $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}-\mathrm{OR}$
3

$\beta$-D-Gulp-( $1 \rightarrow 4$ )- $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}$-OR

$\beta$-D-Glcp-( $1 \rightarrow 4$ )- $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}-\mathrm{OR}$

$\beta$-D-Galp-(1 $\rightarrow 4$ )- $\alpha$-D-Glc $p-1-\mathrm{PO}_{3} \mathrm{H}-\mathrm{OR}$

$\beta$-D-Galp-( $1 \rightarrow 4$ )- $\alpha$-D-Altp-1- $\mathrm{PO}_{3} \mathrm{H}-\mathrm{OR}$
7

$\beta$-D-Galp-( $1 \rightarrow 4$ )- $\alpha$-D-Rhap-1- $\mathrm{PO}_{3} \mathrm{H}-\mathrm{OR}$

$\beta$-D-Fucp-( $1 \rightarrow 4$ )- $\alpha$-D-Manp-1- $\mathrm{PO}_{3} \mathrm{H}$-OR
9

tetrabutylammonium benzoate $\left(\mathrm{Bu}_{4} \mathrm{NOBz}\right)$ in toluene $\left(60^{\circ} \mathrm{C}\right)$ to give the altroside $\mathbf{1 9}(73 \%)$. The D-altro-configuration of the derivative 19 was confirmed by the characteristic values of $J_{2,3}=J_{3,4}=3.0 \mathrm{~Hz}$ in ${ }^{1} \mathrm{H}$ NMR spectrum. Further, compound 19 was converted to the glycosyl acceptor 20 ( $67 \%$ ) by acid hydrolysis followed by selective 6 - $O$-benzoylation with benzoyl cyanide.

Glycosylation of the acceptor $\mathbf{2 0}$ with the bromide $\mathbf{2 1}$ in the presence of silver triflate (AgOTf), silver carbonate and molecular sieves $4 \AA$ in dichloromethane provided the disaccharide 22 in $52 \%$ yield. Hydrogenolysis of compound 22 over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$ afforded a mixture of $\alpha$ - and $\beta$-hemiacetals 24 in the ratio $\alpha: \beta=0.8: 1$ (confirmed by ${ }^{1} \mathrm{H}$ NMR data, see Experimental section). Probably, the mutarotation was facilitated because of unfavourable 1,3-synaxial interaction between $1-\mathrm{OH}$ and 3 -benzoate in the $\alpha$-hemiacetal. The anomeric mixture 24 was converted to the pure $\alpha$-(H-phosphonate) 23 using the same procedure as described for the H-phosphonate 14: i.e., the reaction with triimidazolylphosphine and mild hydrolysis ( $\longrightarrow \mathbf{2 5}$ ) followed by treatment with $\mathrm{H}_{3} \mathrm{PO}_{3}$ in $\mathrm{CH}_{3} \mathrm{CN}$. This produced the H-phosphonate 23 ( $35 \%$ based on the disaccharide 22) along with the recovered hemiacetal 24 (49\%).






Scheme 2 Reagents: i, $\mathrm{Me}_{2} \mathrm{NH}, \mathrm{MeCN}-\mathrm{THF}$; ii, (a) triimidazolylphosphine, MeCN ; (b) $\mathrm{Et}_{3} \mathrm{NHHCO}_{3}$, water ( pH 7 ); iii, $\mathrm{H}_{3} \mathrm{PO}_{3}, \mathrm{MeCN}$; iv, (a) dec-9-en-1-ol, trimethylacetyl chloride, pyridine; (b) $\mathrm{I}_{2}$, pyridinewater; $\mathrm{v}, \mathrm{NaOMe}, \mathrm{MeOH}$.

The hepta- $O$-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-rhamnopyranose 29 (which is a precursor of the phosphodisaccharide 8; Scheme 4) was synthesized using acetobromogalactose 21 and methyl 2,3- $O$-isopropylidene- $\alpha$-D-rhamnopyranoside ${ }^{13}$ 27 as starting materials. Their coupling in the presence of $\mathrm{Hg}(\mathrm{CN})_{2}-\mathrm{HgBr}_{2}$ in acetonitrile-toluene gave the disaccharide 28 ( $74 \%$ ), which was converted to the crystalline heptaacetate 29 in $69 \%$ yield by acid hydrolysis followed by acetolysis/acetylation ${ }^{14}$ with $1.32 \%$ (v/v) $\mathrm{H}_{2} \mathrm{SO}_{4}$ in acetic anhydride.

The hepta- $O$-benzoyl- $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranose 37 (which is a precursor of the phosphodisaccharide 9; Scheme 5) was prepared in $62 \%$ yield by the glycosylation of the D-mannose tetrabenzoate ${ }^{3} 36$ with the $\alpha$-d-fucosyl trichloroacetimidate $\mathbf{3 5}$ in the presence of trimethylsilyl (TMS) triflate. A small proportion of the isomeric $\alpha$-linked disaccharide $40(13 \%$; a precursor of the phosphodisaccharide $\mathbf{1 0}$; Scheme 6) was also isolated from the reaction mixture. The trichloroacetimidate 35 in turn was synthesized from D-fucose by consecutive standard benzoylation $(\longrightarrow 33)$, anomeric deprotection with $\mathrm{Me}_{2} \mathrm{NH}(\longrightarrow \mathbf{3 4} ; 61 \%)$ and the reaction ( $93 \%$ yield) with trichloroacetonitrile in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU). ${ }^{15}$

The $\beta$-configuration of newly formed glycosidic linkages in the disaccharides 22, 28 and 37 followed from the characteristic values of $J_{1^{\prime}, 2^{\prime}}(7.5-8.0 \mathrm{~Hz})$ in ${ }^{1} \mathrm{H}$ NMR spectra. For the $\alpha$-D-fucoside 40 the corresponding value is $J_{1^{\prime}, 2^{\prime}}=3.0 \mathrm{~Hz}$.


Scheme 3 Reagents: i, $\mathrm{BzCN}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{MeCN}$; ii, $\mathrm{Tf}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$-pyridine; iii, $\mathrm{Bu}_{4} \mathrm{NOBz}$, toluene; iv, $80 \% \mathrm{AcOH} ; \mathrm{v}$, $\mathrm{AgOTf}, \mathrm{Ag}_{2} \mathrm{CO}_{3}, \mathrm{MS} 4 \AA$, $\mathrm{CH}_{2} \mathrm{Cl}_{2} ;$ vi, $\mathrm{H}_{2}, \mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$, THF; vii, (a) triimidazolylphosphine, MeCN ; (b) $\mathrm{Et}_{3} \mathrm{NHHCO}_{3}$, water (pH 7); viii, $\mathrm{H}_{3} \mathrm{PO}_{3}, \mathrm{MeCN}$; ix, (a) dec-9-en-1-ol, trimethylacetyl chloride, pyridine; (b) $\mathrm{I}_{2}$, pyridine-water; x , $\mathrm{NaOMe}, \mathrm{MeOH}$.

In contrast to the anomeric de- $O$-acylation of the peracetylated lactose $\mathbf{1 1}$ (see above), similar reaction of the disaccharide heptaacetate 29 and heptabenzoates 37 and 40 with dimethylamine in $\mathrm{CH}_{3} \mathrm{CN}-\mathrm{THF}$ afforded the pure $\alpha$-hemiacetal derivatives 30,38 and 42 ( $66-90 \%$ ), respectively. Compounds 30,38 and 42 were then treated with triimidazolylphosphine followed by mild hydrolysis to produce the $\alpha$-linked glycosyl H-phosphonates 31, 39 and 43, respectively, in 70-97\% yield.
The structures of all the prepared disaccharide H-phosphonates were confirmed by NMR and mass spectrometric data (see Experimental section). For example, signals characteristic of the H-phosphonate group $\left[\delta_{\mathrm{P}} 0.77 ; \delta_{\mathrm{H}} 5.67\right.$ (dd, $J_{1,2} 3.4$, $\left.J_{1, \mathrm{P}} 8.8,1-\mathrm{H}\right), 6.89\left(\mathrm{~d},{ }^{1} J_{\mathrm{H}, \mathrm{P}} 637.8, \mathrm{HP}\right)$ ] were present in the ${ }^{31} \mathrm{P}$ and ${ }^{1} \mathrm{H}$ NMR spectra of the derivative 14 . The $\alpha$-configuration of the D-glucopyranosyl residue followed from the characteristic value of $J_{1,2}$. The main signal in the (electrospray) $\mathrm{ES}(-)$ mass spectrum corresponded to the pseudomolecular ion ( $\mathrm{m} / \mathrm{z}$ 698.9, $\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}$) for the compound. The structures 23, 31, 39 and 43 were established in similar manner apart from that the $\alpha$-configuration of the D-Altp (in compound 23), D-Rhap (in compound 31) and D-Manp (in compounds 39 and 43) residues followed from the characteristic positions of the 3- and $5-\mathrm{H}$ resonances in ${ }^{1} \mathrm{H}$ NMR spectra. The chemical shifts of these signals were close to those of $3-$ and $5-\mathrm{H}$ of the disaccharide derivatives 22 (containing a benzyl 2,3,6-tri- $O$ -benzoyl- $\alpha$-D-altropyranoside moiety), 29 (containing a 1,2,3-tri-O-acetyl- $\alpha$-d-rhamnopyranose moiety) and 37 and 40 (both containing 1,2,3,6-tetra- $O$-benzoyl- $\alpha$-d-mannopyranose moieties), respectively.
The glycosyl H-phosphonates 14, 23, 31, 39 and 43 were converted to the protected phosphodiesters 15, 26, 32, 41 and 44 ( $75-96 \%$ yield), respectively, by their condensation with dec-9-en-1-ol in pyridine in the presence of trimethylacetyl chloride followed by oxidation of the resulting H-phosphonic diesters with iodine in aq. pyridine. The deprotected phosphodi-


Scheme 4 Reagents: i, $\mathrm{Hg}(\mathrm{CN})_{2}, \mathrm{HgBr}_{2}, \mathrm{MeCN}-\mathrm{PhMe}$; ii, (a) aq. TFA, $\mathrm{CHCl}_{3}$; (b) $\mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Ac}_{2} \mathrm{O}$; iii, $\mathrm{Me}_{2} \mathrm{NH}, \mathrm{MeCN}-\mathrm{THF}$; iv, (a) triimidazolylphosphine, MeCN ; (b) $\mathrm{Et}_{3} \mathrm{NHHCO}_{3}$, water ( pH 7 ); v , (a) dec-9-en-1-ol, trimethylacetyl chloride, pyridine; (b) $\mathrm{I}_{2}$, pyridine-water; vi, $\mathrm{NaOMe}, \mathrm{MeOH}$.
saccharides 6-10 were prepared from the derivatives $\mathbf{1 5}, \mathbf{2 6}, \mathbf{3 2}$, 41 and 44, respectively, by de- $O$-acylation with $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ methanolic sodium methoxide in $88-100 \%$ yield.

The structures of the compounds 6-10 and the protected phosphodiesters 15, 26, 32, 41 and 44 were confirmed by NMR and mass spectrometric data. The ${ }^{31} \mathrm{P}$ NMR spectra exhibited single signals $\left[\delta_{\mathrm{P}}\right.$ between -1.35 and -1.96 for the deprotected compounds 6-10 (in $\mathrm{D}_{2} \mathrm{O}$ ) and between -1.67 and -3.12 for the protected phosphodiesters (in $\mathrm{CDCl}_{3}$ )], which are characteristic for glycoside-linked phosphodiesters. ${ }^{3-6,8-10}$ The presence of a $(1 \rightarrow 1)$-phosphodiester linkage at the reducing terminus of each of the disaccharides 6-10 was confirmed by the C-1 and C-2 signals of the corresponding monosaccharide residue and the dec-9-enyl unit in the ${ }^{13} \mathrm{C}$ NMR spectra (Table $1)$. These signals were shifted as a result of the $\alpha$ - and $\beta$-effects of phosphorylation and were coupled with phosphorus (or broadened).

The $\alpha$-configuration of the D-glucopyranosyl phosphate fragments in compounds $\mathbf{6}$ and $\mathbf{1 5}$ was evident from the characteristic values of $J_{1,2}=3.4-3.5 \mathrm{~Hz}$ in the ${ }^{1} \mathrm{H}$ NMR spectra (see Experimental section). The $\alpha$-configuration of the D -altropyranosyl residue in the phosphodisaccharide 7 followed from the characteristic value $\dagger$ of ${ }^{1} J_{\mathrm{C}, \mathrm{H}}=171.3 \mathrm{~Hz}$ for the signal of $\mathrm{C}-1$ and the characteristic position of the C-5 resonance of D-Alt $p$ in the ${ }^{13} \mathrm{C}$ NMR spectrum (Table 1). The chemical shift
$\dagger$ The value of ${ }^{1} J_{\mathrm{C} 1, \mathrm{H} 1} \approx 170 \mathrm{~Hz}$ is typical for $\alpha$-D-derivatives. For the $\beta-$ D-glycosyl residues the value is about 160 Hz : for $\beta$-d-Gal $p$ in compound 7, ${ }^{1} J_{\mathrm{C}^{\prime}, \mathrm{H1}^{\prime}}=162.5 \mathrm{~Hz}$ (Table 1) (see also refs. 3, 4, 9 and 16).


Scheme 5 Reagents: i, $\mathrm{Me}_{2} \mathrm{NH}, \mathrm{MeCN}-\mathrm{THF}$; ii, $\mathrm{CCl}_{3} \mathrm{CN}$, DBU, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; iii, TMS triflate, MS $4 \AA, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; iv, (a) triimidazolylphosphine, MeCN ; (b) $\mathrm{Et}_{3} \mathrm{NHHCO}_{3}$, water ( pH 7 ); v, (a) dec-9-en-1-ol, trimethylacetyl chloride, pyridine; (b) $\mathrm{I}_{2}$, pyridine-water; vi, NaOMe , MeOH .

iii $\downarrow$

10

Scheme 6 Reagents: i, $\mathrm{Me}_{2} \mathrm{NH}, \mathrm{MeCN}-\mathrm{THF}$; ii, (a) triimidazolylphosphine, MeCN ; (b) $\mathrm{Et}_{3} \mathrm{NHHCO}_{3}$, water (pH 7); iii, (a) dec-9-en-1ol, trimethylacetyl chloride, pyridine; (b) $\mathrm{I}_{2}$, pyridine-water; iv, $\mathrm{NaOMe}, \mathrm{MeOH}$.

Table $1{ }^{13} \mathrm{C}$ and ${ }^{31} \mathrm{P}$ NMR data $\left[\delta_{\mathrm{C}}\right.$ and $\delta_{\mathrm{P}}$ in ppm; $J_{\mathrm{C}, \mathrm{P}}$ and $J_{\mathrm{C}, \mathrm{H}}$ in Hz ; spectra recorded in $\mathrm{D}_{2} \mathrm{O}$ ] and $\mathrm{ESMS}(-)$ data ( $\mathrm{m} / \mathrm{z}$ ) for the phosphodisaccharides 6-10

| Residue | Atom | $6^{a}$ | $7^{a}$ | $8^{\text {b }}$ | $9^{a}$ | $10^{a}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Dec-9-enyl | $\mathrm{OCH}_{2} \mathrm{CH}_{2}$ | 67.20 d | 67.82d | 67.76 d | 67.45 br | 67.01 br |
|  |  | $J_{\mathrm{C}, \mathrm{P}} 4.0$ | $J_{\text {C,P }} \approx 6$ | $J_{\text {C,P }} \approx 6$ |  |  |
|  | $\mathrm{OCH}_{2} \mathrm{CH}_{2}$ | 31.15 d | 30.96d | 30.87 d | 30.95 br | 31.03 d |
|  |  | $J_{\mathrm{C}, \mathrm{P}} 9.0$ | $J_{\text {C,P }} 8.8$ | $J_{\text {C,P }} 5.9$ |  | $J_{\text {C,P }} 8.3$ |
|  | $-\mathrm{CH}=$ | 140.32 | 141.64 | 141.54 | 140.87 | 141.52 |
|  | $=\mathrm{CH}_{2}$ | 115.10 | 115.09 | 115.00 | 115.00 | 114.98 |
| Aldose | C-1 | $95.71 \mathrm{~d}$ | 96.76 br | 96.70 br | 96.69 br | $96.78 \mathrm{~d}$ |
|  |  | $J_{\mathrm{C}, \mathrm{P}} 6.9$ |  |  |  | $J_{\mathrm{C}, \mathrm{P}} 5.8$ |
|  |  |  | $J_{\text {C,H }} 171.3$ | $J_{\text {C,H }} 169.7$ | $J_{\mathrm{C}, \mathrm{H}} 171.0$ | $J_{\text {C,H }} 170.5$ |
|  | C-2 | 72.10 d | 71.54 d | 71.18 d | 70.89d | 71.29 d |
|  |  | $J_{\text {C,P }} 7.7$ | $J_{\mathrm{C}, \mathrm{P}} 10.0$ | $J_{\mathrm{C}, \mathrm{P}} 6.9$ | $J_{\mathrm{C}, \mathrm{P}} 7.2$ | $J_{\mathrm{C}, \mathrm{P}} 7.5$ |
|  | C-3 | 72.23 | 70.95 | 69.61 | 69.76 | 70.54 |
|  | C-4 | 78.77 | 74.38 | 82.62 | 77.20 | 76.98 |
|  | C-5 | 72.39 | 69.78 | 69.44 | 73.12 | 73.30 |
|  | C-6 | 60.65 | 61.53 | 17.82 | 61.10 | 61.30 |
| Aldose ${ }^{\prime}$ | C-1' | 103.85 | 105.04 | 104.27 | 103.87 | 102.08 |
|  |  |  | $J_{\text {C,H }} 162.5$ | $J_{\text {C,H }} 161.0$ | $J_{\mathrm{C}, \mathrm{H}} 160.5$ | $J_{\text {C,H }} 171.0$ |
|  | C-2' | 71.83 | $71.96$ | 72.09 | 71.45 | 69.64 |
|  | C-3' | 73.58 | 73.67 | 73.61 | 73.71 | 71.23 |
|  | C-4' | 69.51 | 69.96 | 69.72 | 72.12 | 72.57 |
|  | C-5' | 76.25 | 76.21 | 76.37 | 72.01 | 68.22 |
|  | C-6' | 61.93 | 62.20 | 62.18 | 16.34 | 16.42 |
| Phosphate | P | -1.96 | $-1.35$ | $-1.50$ | -1.41 | $-1.66$ |
|  | $m / z^{c}$ | 559.34 | 558.90 | 543.25 | 543.10 | 543.10 |

${ }^{a}$ Additional signals of $\mathrm{Et}_{3} \mathrm{NH}^{+}\left[\delta_{\mathrm{C}} 9.20-9.37\left(\mathrm{CH}_{3}\right)\right.$ and $\left.\delta_{\mathrm{C}} 47.41-47.63\left(\mathrm{CH}_{2}\right)\right]$ were present. ${ }^{a, b}$ Additional signals of $\mathrm{CC} \mathrm{H}_{2} \mathrm{C}\left[\delta_{\mathrm{C}} 25.95-26.26,29.19-\right.$ 30.09 and $34.16-34.44$ ] were present. ${ }^{c}$ Corresponds to the pseudomolecular ions $\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right.$ ] ${ }^{-}$. For compounds 6 and 7 (triethylammonium salt), $\mathrm{C}_{28} \mathrm{H}_{56} \mathrm{NO}_{14} \mathrm{P}$ requires $M, 661.34$ (expected $m / z, 559.14$ ); for compounds $\mathbf{8}-\mathbf{1 0}$ (triethylammonium salt), $\mathrm{C}_{28} \mathrm{H}_{56} \mathrm{NO}_{13} \mathrm{P}$ requires $M$, 645.35 (expected $m / z, 534.15$ ).
of the C-5 signal ( $\delta_{\mathrm{C}} 69.78$ ) is fairly close to that of C-5 ( $\delta_{\mathrm{C}}$ $70.00) \ddagger$ of methyl $\alpha$-D-altropyranoside. ${ }^{17}$

The $\alpha$-configuration of the D-mannopyranosyl phosphate fragments in compounds 9 and 10 and of the D-rhamnopyranosyl phosphate in compound $\mathbf{8}$ was confirmed by 1) the characteristic valuest of ${ }^{1} J_{\mathrm{C}, \mathrm{H}}$ for the signals of $\mathrm{C}-1$ and 2 ) the characteristic positions of the C-3 and C-5 resonances of D-Man $p$ and D-Rhap residues, respectively, in the ${ }^{13} \mathrm{C}$ NMR spectra (see Table 1). The chemical shifts of the signals of C-3 and -5 of D-Manp and C-3 of D-Rhap (i.e., 6-deoxy-Dmannose) are close to those of C-3 and C-5 of $\alpha$-D-mannopyranosyl phosphate ${ }^{18}$ taking into account the influence of the glycosyl substituents at position 4. The chemical shift of C-5 resonance ( $\delta_{\mathrm{C}} 69.44$ ) of D-Rhap in compound $\mathbf{8}$ is very close to that of C-5 ( $\left.\delta_{\mathrm{C}} 69.40\right) \S$ of methyl $\alpha$-D-rhamnopyranoside. ${ }^{17}$

The $\alpha$-configuration of the glycosyl phosphate linkages in the protected derivatives $26,32,41$ and 44 followed from the characteristic positions of $1-, 3$ - and $5-\mathrm{H}$ resonances in their ${ }^{1} \mathrm{H}$ NMR spectra (see Experimental section).
The molecular masses of the phosphodiesters 6-10, 15, 26, 32, 41 and 44 were confirmed by electrospray mass spectrometry. The signals in the ES(-) mass spectra corresponded to the pseudomolecular ions for the disaccharide phosphates (see Table 1 and Experimental section). A biochemical evaluation of compounds 6-10 will be published elsewhere ${ }^{19}$ in due course.

## Experimental

## General procedures

Optical rotations were measured with a Perkin-Elmer 141 polarimeter; $[a]_{\mathrm{D}}$-values are given in units of $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$. NMR spectra ( ${ }^{1} \mathrm{H}$ at 200 and $500 \mathrm{MHz},{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ at 50.3 and 125 MHz , and ${ }^{31} \mathrm{P}\left\{{ }^{1} \mathrm{H}\right\}$ at 81 and 202.5 MHz ) were recorded with

[^0]Bruker AM-200 and AM-500 spectrometers for solutions in $\mathrm{CDCl}_{3}$, unless otherwise indicated. Chemical shifts ( $\delta$ in ppm ) are given relative to those for $\mathrm{Me}_{4} \mathrm{Si}$ (for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ ) and external aq. $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}$ (for ${ }^{31} \mathrm{P}$ ); $J$-values are given in Hz . ES mass spectra were recorded with a Micromass Quattro system (Micromass Biotech, UK). TLC was performed on Kieselgel $60 \mathrm{~F}_{254}$ (Merck) with $A$, toluene-ethyl acetate (95:5); B, toluene-ethyl acetate (9:1); $C$, toluene-ethyl acetate (7:3); $D$, toluene-ethyl acetate ( $3: 7$ ); $E$, dichloromethane-methanol (95:5); $F$, chloroform-methanol (8:2); and $G$, chloroform-methanol-water ( $10: 10: 3$ ) as developers and detection under UV light or by charring with sulfuric acid-water-ethanol (15:85:5). Flash-column chromatography (FCC) was performed on Kieselgel 60 ( $0.040-0.063 \mathrm{~mm}$ ) (Merck). Dichloromethane, acetonitrile and toluene were freshly distilled from $\mathrm{CaH}_{2}$. Solutions worked up were concentrated under reduced pressure at $<40^{\circ} \mathrm{C}$.

## 2,3,4,6-Tetra- $O$-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -acetyl- $\alpha, \beta$-d-glucopyranose 12

To a solution of 2,3,4,6-tetra- $O$-acetyl- $\beta$-D-galactopyranosyl$(1 \rightarrow 4)$-1,2,3,6-tetra- $O$-acetyl- $\alpha, \beta$-d-glucopyranose $11(1 \mathrm{~g}, 1.47$ mmol ) [prepared by standard acetylation of $\alpha$-lactose with $\mathrm{Ac}_{2} \mathrm{O}$ in pyridine at $0^{\circ} \mathrm{C} ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ (inter alia) $1.94,1.98$, 2.10, 2.13 and $2.15(15 \mathrm{H}, 5 \times \mathrm{s}, 5 \times \mathrm{Ac}), 2.03(9 \mathrm{H}, \mathrm{s}, 3 \times \mathrm{Ac})$, $3.79\left(\mathrm{t}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}^{+}\right), 3.86\left(1 \mathrm{H}, \mathrm{dt}, J_{5^{\prime}, 6^{\prime}} 6.6,5^{\prime}-\mathrm{H}\right)$, 3.93-4.17 $\left(4 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}, 6-\mathrm{H}^{\mathrm{a}}\right.$ and $\left.6^{\prime}-\mathrm{H}_{2}\right), 4.42\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.6\right.$, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.4,6-\mathrm{H}^{\mathrm{b}}\right), 4.45\left(\mathrm{~d}, J_{1^{\prime}, 2^{\prime}} 7.9,1^{\prime}-\mathrm{H}^{d}\right), 4.54\left(\mathrm{~d}, J_{1^{\prime}, 2^{\prime}} 8.2\right.$, $\left.1^{\prime}-\mathrm{H}^{\beta}\right), 4.94\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4} 3.4,3^{\prime}-\mathrm{H}\right), 4.98$ (dd, $\left.J_{2,3} 9.9,2-\mathrm{H}^{\alpha}\right)$, $5.10\left(\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.6,2^{\prime}-\mathrm{H}^{\text {a }}\right.$ ), $5.33\left(1 \mathrm{H}, \mathrm{dd}, J_{4^{\prime}, 5^{\prime}} 0.5,4^{\prime}-\mathrm{H}\right), 5.44$ (dd, $\left.3-\mathrm{H}^{\alpha}\right), 5.65\left(\mathrm{~d}, J_{1,2} 7.1,1-\mathrm{H}^{\beta}\right)$ and $6.22\left(\mathrm{~d}, J_{1,2} 3.6,1-\mathrm{H}^{\alpha}\right)$; $\alpha: \beta \approx 7: 1]$ in acetonitrile $\left(6 \mathrm{~cm}^{3}\right)$ was added $2 \mathrm{~mol} \mathrm{dm}^{-3}$ $\mathrm{Me}_{2} \mathrm{NH}$ in THF $\left(4 \mathrm{~cm}^{3} ; 7.96 \mathrm{mmol}\right)$ and the mixture was kept at rt with monitoring by TLC (solvent $D$ ). After 4-9 h the mixture was concentrated to dryness and acetonitrile was evaporated off from the residue. FCC [ethyl acetate-toluene, $(2: 8) \longrightarrow$ (8:2)] of the residue gave the disaccharide $a, \beta$-hemiacetal 12
$(0.779 \mathrm{~g}, 83 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{25}+35.2$ (c 1.06 , $\mathrm{CHCl}_{3}$ ) (Found: C, 48.8; H, 5.6. $\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{O}_{18}$ requires C, 49.1; H, $5.7 \%) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ (inter alia) $1.96,2.03,2.04,2.05,2.07,2.12$ and $2.15(21 \mathrm{H}, 7 \times \mathrm{s}, 7 \times \mathrm{Ac}), 3.75\left(\mathrm{dd}, J_{4,5} 9.3,4-\mathrm{H}^{\alpha}\right), 3.86(1$ $\left.\mathrm{H}, \mathrm{dt}, J_{5^{\prime}, 6^{\prime}} 6.3,5^{\prime}-\mathrm{H}\right), 4.00-4.22\left(4 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}, 6-\mathrm{H}^{\mathrm{a}}\right.$ and $6^{\prime}-\mathrm{H}_{2}$ ), 4.47 (d, $\left.J_{1^{\prime}, 2^{\prime}} 7.7,1^{\prime}-\mathrm{H}^{\beta}\right), 4.48\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 3.4, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.2\right.$, $\left.6-\mathrm{H}^{\mathrm{b}}\right), 4.49\left(\mathrm{~d}, J_{1^{\prime}, 2^{\prime}} 7.9,1^{\prime}-\mathrm{H}^{\mathrm{d}}\right), 4.76\left(\mathrm{~m}, 1-\mathrm{and} 2-\mathrm{H}^{\beta}\right), 4.81(\mathrm{dd}$, $\left.2-\mathrm{H}^{\mathrm{o}}\right), 4.94\left(1 \mathrm{H}\right.$, dd, $\left.J_{3^{\prime}, 4} 3.2,3^{\prime}-\mathrm{H}\right), 5.09\left(\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.6,2^{\prime}-\mathrm{H}^{\beta}\right)$, $5.11\left(\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.5,2^{\prime}-\mathrm{H}^{u}\right), 5.22\left(\mathrm{t}, J_{2,3}=J_{3,4}=9.3,3-\mathrm{H}^{\beta}\right), 5.34$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{4^{\prime}, 5^{\prime}} 0.5,4^{\prime}-\mathrm{H}\right), 5.36\left(\mathrm{~d}, J_{1,2} 3.4,1-\mathrm{H}^{\omega}\right)$ and $5.51(\mathrm{t}$, $\left.J_{2,3}=J_{3,4}=9.7,3-\mathrm{H}^{\alpha}\right) ; \alpha: \beta=4: 1$.

2,3,4,6-Tetra- $O$-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -acetyl- $\alpha$-D-glucopyranosyl hydrogenphosphonate, triethylammonium salt 14

To a stirred solution of imidazole ( $0.65 \mathrm{~g}, 9.56 \mathrm{mmol}$ ) in acetonitrile $\left(10 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$ was added phosphorus trichloride ( $\left.0.25 \mathrm{~cm}^{3}, 2.87 \mathrm{mmol}\right)$ followed by $\mathrm{Et}_{3} \mathrm{~N}\left(1.4 \mathrm{~cm}^{3}, 10.04 \mathrm{mmol}\right)$. The mixture was stirred for 20 min , after which a solution of compound $\mathbf{1 2}(0.304 \mathrm{~g}, 0.478 \mathrm{mmol})$ in $\mathrm{MeCN}\left(10 \mathrm{~cm}^{3}\right)$ was added dropwise over a period of $10-15 \mathrm{~min}$ at $0^{\circ} \mathrm{C}$. The mixture was stirred at rt for $30-40 \mathrm{~min}$ and quenched with 1 mol $\mathrm{dm}^{-3}$ triethylammonium (TEA) hydrogen carbonate ( pH 7 ; $4 \mathrm{~cm}^{3}$ ). The clear solution was stirred for $15 \mathrm{~min}, \mathrm{CH}_{2} \mathrm{Cl}_{2}(100$ $\mathrm{cm}^{3}$ ) was added and the organic layer was washed in turn with ice-cold water $\left(2 \times 40 \mathrm{~cm}^{3}\right)$ and cold $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ TEA hydrogen carbonate ( $2 \times 40 \mathrm{~cm}^{3}$ ), dried by filtration through cotton wool, and concentrated to give the $\alpha, \beta$-( H -phosphonate) 13, $\delta_{\mathrm{P}} 0.43\left(\mathrm{P}^{\alpha}\right)$ and $1.20\left(\mathrm{P}^{\beta}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ (inter alia) $5.20\left(\mathrm{dd}, J_{1,2}\right.$ 7.7, $\left.J_{1, \mathrm{P}} 9.1,1-\mathrm{H}^{\mathrm{\beta}}\right), 5.67\left(\mathrm{dd}, J_{1,2} 3.4, J_{1, \mathrm{P}} 8.8,1-\mathrm{H}^{\alpha}\right), 6.88\left(\mathrm{~d},{ }^{1} J_{\mathrm{H}, \mathrm{P}}\right.$ 644.0, $\mathrm{H}-\mathrm{P}^{\beta}$ ) and 6.91 (d, ${ }^{1} J_{\mathrm{H}, \mathrm{P}} 637.8, \mathrm{H}-\mathrm{P}^{\alpha}$ ); $\alpha: \beta=4: 1$.

The residue was dissolved in $\mathrm{CH}_{3} \mathrm{CN}\left(15 \mathrm{~cm}^{3}\right)$ and anhydrous $\mathrm{H}_{3} \mathrm{PO}_{3}(0.67 \mathrm{~g}, 8.17 \mathrm{mmol})$ was added. The mixture was stirred at rt for 19 h , then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(100 \mathrm{~cm}^{3}\right)$ and washed successively with cold saturated aq. $\mathrm{NaHCO}_{3}\left(2 \times 40 \mathrm{~cm}^{3}\right)$ and cold $0.5 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{aq}$. TEA hydrogen carbonate $\left(2 \times 40 \mathrm{~cm}^{3}\right)$. The organic phase (containing the hemiacetal 12) was discarded. The aqueous washings were then combined, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(4 \times 40 \mathrm{~cm}^{3}\right)$. The combined organic washings were dried by filtration through cotton wool, and concentrated to produce the $\alpha$-hydrogenphosphonate 14 ( 0.184 $\mathrm{g}, 48 \%$ ) as a chromatographically homogeneous amorphous solid, $[a]_{\mathrm{D}}^{26}+41.8\left(c 0.97, \mathrm{CHCl}_{3}\right)$; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.32(9 \mathrm{H}, \mathrm{t}$, $\left.3 \times M e \mathrm{CH}_{2}\right), 1.92,2.00,2.02,2.07$ and $2.10(15 \mathrm{H}, 5 \times \mathrm{s}$, $5 \times \mathrm{Ac}), 1.99(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Ac}), 3.04\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.74$ $\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 3.84\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6^{\prime}} 6.7,5^{\prime}-\mathrm{H}\right)$, 3.97-4.20 $\left(4 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}, 6-\mathrm{H}^{\mathrm{a}}\right.$ and $\left.6^{\prime}-\mathrm{H}_{2}\right), 4.41\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.7\right.$, $\left.1^{\prime}-\mathrm{H}\right), 4.42\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 10.9,6-\mathrm{H}^{\mathrm{b}}\right), 4.84\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 3.4\right.$, $2-\mathrm{H}), 4.90\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.3,3^{\prime}-\mathrm{H}\right), 5.06\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.3\right.$, $\left.2^{\prime}-\mathrm{H}\right), 5.30\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right), 5.44\left(1 \mathrm{H}, \mathrm{t}, J_{2,3} 9.6,3-\mathrm{H}\right), 5.67(1 \mathrm{H}$, $\left.\mathrm{dd}, J_{1, \mathrm{P}} 8.8,1-\mathrm{H}\right)$ and $6.89\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{H}, \mathrm{P}} 637.8, \mathrm{HP}\right) ; \delta_{\mathrm{P}} 0.77$; ESMS(-) data: $m / z 698.9\left(100 \% \text {, [ } \mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}$) (expected $m / z$, 699.08. $\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{NO}_{20} \mathrm{P}$ requires $M, 801.28$ ).

Dec-9-enyl 2,3,4,6-tetra- $O$-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- O-acetyl- $\alpha$-d-glucopyranosyl phosphate, triethylammonium salt 15

A mixture of the H -phosphonate $\mathbf{1 4}(126 \mathrm{mg}, 0.16 \mathrm{mmol})$ and dec-9-en-1-ol ( $0.056 \mathrm{~cm}^{3}, 0.31 \mathrm{mmol}$ ) was dried by evaporation of pyridine $\left(3 \times 2 \mathrm{~cm}^{3}\right)$ therefrom. The residue was dissolved in pyridine ( $1 \mathrm{~cm}^{3}$ ), trimethylacetyl chloride $\left(0.048 \mathrm{~cm}^{3}, 0.39\right.$ mmol ) was added, and the mixture was stirred at rt for $10-15$ min , whereafter a freshly prepared solution of iodine ( 80 mg , 0.314 mmol ) in pyridine-water ( $95: 5 ; 2 \mathrm{~cm}^{3}$ ) was added. After $30 \mathrm{~min}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added and the solution was washed successively with ice-cold $1 \mathrm{~mol} \mathrm{dm}^{-3}$ aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ and cold 0.5 mol $\mathrm{dm}^{-3}$ aq. TEA hydrogen carbonate, dried by filtration through cotton wool, and concentrated. FCC $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}\right.$, (99:0:1) $\longrightarrow(89: 10: 1)]$ of the residue gave the phospho-
diester $15(144 \mathrm{mg}, 96 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{25}+37$ (c $\left.0.96, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.20-1.32\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right)$, $1.29\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.57\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)$, 1.93, 2.01, 2.02, 2.09 and $2.12(15 \mathrm{H}, 5 \times \mathrm{s}, 5 \times \mathrm{Ac}), 2.00(8 \mathrm{H}, \mathrm{s}$, $2 \times \mathrm{Ac}$ and $\left.\mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 3.03\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.76$ $\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 3.83\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6^{\prime}} 6.7,5^{\prime}-\mathrm{H}\right), 3.85$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.98-4.12\left(3 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{a}}\right.$ and $\left.6^{\prime}-\mathrm{H}_{2}\right), 4.17$ $\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{a}} 2.4,5-\mathrm{H}\right), 4.41\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{2}} 7.7,1^{\prime}-\mathrm{H}\right), 4.45(1 \mathrm{H}$, dd, $\left.J_{5,6 \mathrm{~b}} 1.3, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.0,6-\mathrm{H}^{\mathrm{b}}\right), 4.83\left(1 \mathrm{H}\right.$, ddd, $J_{1,2} 3.4, J_{2, \mathrm{P}} 1.9$, $2-\mathrm{H}), 4.90\left(1 \mathrm{H}, \mathrm{dd}, 3^{\prime}-\mathrm{H}\right), 4.86\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.4,{ }^{3} J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}} 10.3\right.$, $H C H=\mathrm{CH}), 4.95\left(1 \mathrm{H}, \mathrm{dd},{ }^{3} J_{\mathrm{H}, \mathrm{H}-E} 17.0, \mathrm{HC} H=\mathrm{CH}\right), 5.07(1 \mathrm{H}$, $\left.\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.4,2^{\prime}-\mathrm{H}\right), 5.31\left(1 \mathrm{H}, \mathrm{d}, J_{3^{\prime}, 4^{\prime}} 3.2,4^{\prime}-\mathrm{H}\right), 5.45(1 \mathrm{H}, \mathrm{t}$, $\left.J_{2,3} 9.6,3-\mathrm{H}\right), 5.63\left(1 \mathrm{H}, \mathrm{dd}, J_{1, \mathrm{P}} 8.0,1-\mathrm{H}\right)$ and $5.77(1 \mathrm{H}$, ddt, $\left.J_{\mathrm{H}, \mathrm{CH}}^{2}, 6.7, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{P}}-1.67$; ESMS( - ): m/z 853.0 $\left(100^{\circ} \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$(expected $m / z$, 853.22. $\mathrm{C}_{42} \mathrm{H}_{70} \mathrm{NO}_{21} \mathrm{P}$ requires $M, 955.42$ ).

## Dec-9-enyl $\boldsymbol{\beta}$-D-galactopyranosyl-( $\mathbf{1} \rightarrow \mathbf{4}$ )- $\alpha$-D-glucopyranosyl phosphate, triethylammonium salt 6

To a solution of compound $\mathbf{1 5}(138 \mathrm{mg})$ in $\mathrm{MeOH}\left(15 \mathrm{~cm}^{3}\right)$ was added $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ methanolic $\mathrm{NaOMe}\left(1.7 \mathrm{~cm}^{3}\right)$. The mixture (now $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ in NaOMe ) was kept at room temperature for 1 h , whereafter it was deionized with Dowex $50 \mathrm{~W}-\mathrm{X} 4\left(\mathrm{H}^{+}\right)$ resin, filtered, and immediately neutralized with $\mathrm{Et}_{3} \mathrm{~N}$. The solution was concentrated and methanol was evaporated off from the residue. The phosphodiester $\mathbf{6}(96 \mathrm{mg}, 100 \%)$ was thereby obtained as an amorphous solid, $[a]_{\mathrm{D}}^{25}+54.5$ (c 1, MeOH); $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) $1.16\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.19$ $\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.48\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.91$ ( $2 \mathrm{H}, \mathrm{dt}, \mathrm{J} 6.6, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), $3.06\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 4.34$ $\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.6,1^{\prime}-\mathrm{H}\right), 5.33\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 3.5, J_{1, \mathrm{P}} 7.1,1-\mathrm{H}\right)$ and $5.71\left(1 \mathrm{H}, \mathrm{ddt}, J_{\mathrm{H}, \mathrm{CH}}^{2} 2.6, J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}} 10.2, J_{\mathrm{H}, \mathrm{H}-E} 17.0, \mathrm{CH}_{2}-\right.$ $\left.\mathrm{C} H=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}, \delta_{\mathrm{P}}$ and ESMS(-) data: see Table 1.

## Benzyl 2-O-benzoyl-4,6-O-isopropylidene- $\alpha$-D-mannopyranoside 17

To a stirred solution of benzyl 4,6-O-isopropylidene- $\alpha$-Dmannopyranoside 16 ( $3.1 \mathrm{~g}, 10 \mathrm{mmol}$ ) [prepared from benzyl $\alpha$-D-mannopyranoside and 2-methoxypropene in $85 \%$ yield, $[\alpha]_{D}^{25}$ $+85\left(c 1, \mathrm{CHCl}_{3}\right), R_{\mathrm{f}} 0.3$ (solvent $E$ ) (Found: C, 61.8; H, 7.2. $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{6}$ requires $\mathrm{C}, 61.9 ; \mathrm{H}, 7.1 \%$ ) as described for the preparation of methyl 4,6-O-isopropylidene- $\alpha$-d-mannopyranoside ${ }^{11}$ ] and $\mathrm{BzCN}(1.57 \mathrm{~g}, 12 \mathrm{mmol})$ in acetonitrile $\left(20 \mathrm{~cm}^{3}\right)$ was added $\mathrm{Et}_{3} \mathrm{~N}\left(0.025 \mathrm{~cm}^{3}\right)$. After 30 min , methanol was added, the reaction mixture was concentrated, and toluene was evaporated off from the residue. FCC (solvent $A$ ) gave the monobenzoate 17 $(3.15 \mathrm{~g}, 76 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{25}+48\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}}$ 0.2 (solvent $B$ ) (Found: C, $66.25 ; \mathrm{H}, 6.3 . \mathrm{C}_{23} \mathrm{H}_{26} \mathrm{O}_{7}$ requires C, $66.65 ; \mathrm{H}, 6.3 \%) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.49$ and $1.61(6 \mathrm{H}, 2 \times \mathrm{s}$, $2 \times \mathrm{Me}), 3.73-3.93\left(3 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}_{2}\right), 4.07(1 \mathrm{H}, \mathrm{t}$, $\left.J_{3,4}=J_{4,5}=9.0,4-\mathrm{H}\right), 4.26\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.4,3-\mathrm{H}\right), 4.53$ and 4.73 $\left(2{ }^{2} \mathrm{H}, \mathrm{AB}\right.$ q, $\left.J 11.7, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.00\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.3,1-\mathrm{H}\right), 5.50$ ( $1 \mathrm{H}, \mathrm{dd}, 2-\mathrm{H}$ ) and 7.15-8.15 ( $10 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{Ph}$ ).

## Benzyl 2,3-di- $O$-benzoyl-4,6-O-isopropylidene- $\alpha$-D-altropyranoside 19

Triflic anhydride ( $2.05 \mathrm{~cm}^{3}, 12.2 \mathrm{mmol}$ ) was added dropwise to a cooled $\left(0^{\circ} \mathrm{C}\right)$ stirred solution of compound $17(2.53 \mathrm{~g}, 6.11$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(50 \mathrm{~cm}^{3}\right)$ containing pyridine ( $3.85 \mathrm{~cm}^{3}, 48.9$ mmol ), and then the reaction mixture was allowed to warm to rt. After 1 h , the mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, washed successively with ice-cold $0.1 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{HCl}$, ice-cold saturated aq. $\mathrm{NaHCO}_{3}$ and water, and dried by filtration through cotton wool. The filtrate was concentrated to dryness and toluene was evaporated off from the residue to produce the triflate $\mathbf{1 8}\left[R_{\mathrm{f}} 0.5\right.$ (solvent $B), \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.45$ and $1.58(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{Me})$, $3.83-3.93\left(3 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}_{2}\right), 4.28\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.1\right.$, $4-\mathrm{H}), 4.57$ and $4.72\left(2 \mathrm{H}, \mathrm{AB} \mathrm{q}, J 11.7, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.02(1 \mathrm{H}, \mathrm{d}$,
$\left.J_{1,2} 1.1,1-\mathrm{H}\right), 5.25\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.6,3-\mathrm{H}\right), 5.65(1 \mathrm{H}, \mathrm{dd}, 2-\mathrm{H})$ and 7.10-8.10 $(10 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{Ph})$ ].

A solution of tetrabutylammonium benzoate $(3.63 \mathrm{~g}, 10$ mmol ; dried beforehand by evaporation of anhydrous toluene therefrom) in toluene ( $20 \mathrm{~cm}^{3}$ ) was added to a solution of the triflate $\mathbf{1 8}$ in the same solvent $\left(30 \mathrm{~cm}^{3}\right)$. The reaction mixture was stirred at $60^{\circ} \mathrm{C}$ for 7 h , then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried by filtration through cotton wool and concentrated. FCC (solvent A) gave the altroside $19(2.3 \mathrm{~g}, 73 \%)$, $\mathrm{mp} 142-144{ }^{\circ} \mathrm{C}$ (from diethyl ether-hexane); $[a]_{\mathrm{D}}^{25}+16\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.5$ (solvent $B$ ) (Found: C, 69.8; H, 5.9. $\mathrm{C}_{30} \mathrm{H}_{30} \mathrm{O}_{8}$ requires C, 69.5; H, $5.8 \%$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.35$ and $1.61(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{Me}), 3.90(1 \mathrm{H}, \mathrm{t}$, $\left.J_{5,6 \mathrm{a}}=J_{6 \mathrm{a}, 6 \mathrm{~b}}=9.6,6-\mathrm{H}^{\mathrm{a}}\right), 3.98\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 5.7,6-\mathrm{H}^{\mathrm{b}}\right), 4.26$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.6,4-\mathrm{H}\right), 4.43(1 \mathrm{H}, \mathrm{dt}, 5-\mathrm{H}), 4.53$ and $4.82(2 \mathrm{H}$, $\left.\mathrm{AB} \mathrm{q}, J 11.1, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.03\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.1,1-\mathrm{H}\right), 5.39(1 \mathrm{H}, \mathrm{dd}$, $2-\mathrm{H}), 5.55\left(1 \mathrm{H}, \mathrm{t}, J_{2,3}=J_{3,4}=3.0,3-\mathrm{H}\right)$ and $7.15-8.15(15 \mathrm{H}, \mathrm{m}$, $3 \times \mathrm{Ph}$ ).

## Benzyl 2,3,6-tri- $O$-benzoyl- $\boldsymbol{\alpha}$-D-altropyranoside 20

A solution of the altroside $\mathbf{1 9}(2.49 \mathrm{~g}, 4.8 \mathrm{mmol})$ in $80 \% \mathrm{aq}$. acetic acid $\left(50 \mathrm{~cm}^{3}\right)$ was heated at $60{ }^{\circ} \mathrm{C}$ for 1 h , whereafter the mixture was concentrated and toluene was twice evaporated off from the residue. The residue was dissolved in acetonitrile ( 50 $\left.\mathrm{cm}^{3}\right)$ and $\mathrm{BzCN}(0.63 \mathrm{~g}, 4.82 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}\left(0.025 \mathrm{~cm}^{3}\right)$ were added to the solution. After 30 min , methanol was added, the reaction mixture was concentrated, and toluene was evaporated off from the residue. FCC (solvent $A$ ) gave the tribenzoate 20 $(1.87 \mathrm{~g}, 67 \%), \mathrm{mp} 140-142^{\circ} \mathrm{C}$ (from diethyl ether-hexane); $[a]_{\mathrm{D}}^{25}-6.5\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.25$ (solvent $B$ ) (Found: C, $70.4 ; \mathrm{H}$, 5.1. $\mathrm{C}_{34} \mathrm{H}_{30} \mathrm{O}_{9}$ requires $\mathrm{C}, 70.1 ; \mathrm{H}, 5.2 \%$ ); $\delta_{\mathrm{H}}$ ( 200 MHz ; $\left.\mathrm{CDCl}_{3}+\mathrm{D}_{2} \mathrm{O}\right) 4.28\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.7,4-\mathrm{H}\right), 4.48\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}}\right.$ $2.3,5-\mathrm{H}), 4.58$ and $4.83\left(2 \mathrm{H}, \mathrm{AB} \mathrm{q}, J 10.8, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.66(1 \mathrm{H}$, dd, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.0,6-\mathrm{H}^{\mathrm{a}}\right), 4.78\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 4.0,6-\mathrm{H}^{\mathrm{b}}\right), 5.10(1 \mathrm{H}, \mathrm{d}$, $\left.J_{1,2} 1.0,1-\mathrm{H}\right), 5.42(1 \mathrm{H}, \mathrm{dd}, 2-\mathrm{H}), 5.62\left(1 \mathrm{H}, \mathrm{t}, J_{2,3}=J_{3,4}=3.2\right.$, 3-H) and 7.15-8.15 ( $20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph}$ ).

## Benzyl 2,3,4,6-tetra- $O$-acetyl- $\beta$-d-galactopyranosyl-(1 $\rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-dl-altropyranoside 22

A solution of acetobromogalactose $21(1.03 \mathrm{~g}, 2.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(15 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred mixture of the altroside $20(0.73 \mathrm{~g}, 1.25 \mathrm{mmol}), \mathrm{Ag}_{2} \mathrm{CO}_{3}(1.37 \mathrm{~g}, 5.0 \mathrm{mmol})$, AgOTf ( $0.64 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) and freshly activated molecular sieves $4 \AA$ (powder, 5 g ) in boiling dichloromethane ( $30 \mathrm{~cm}^{3}$ ). The reaction mixture was stirred under reflux for 2.5 h and then at rt for 20 h . The solids were filtered off and the filtrate was concentrated. FCC [diethyl ether-hexane, $(1: 1) \longrightarrow(2: 1)$ ] gave the disaccharide $22(0.59 \mathrm{~g}, 52 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{25}+12\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.45$ (solvent $C$ ) (Found: C, 63.1; H, 5.4. $\mathrm{C}_{48} \mathrm{H}_{48} \mathrm{O}_{18}$ requires C, $63.15 ; \mathrm{H}, 5.3 \%$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.91$ $(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Ac}), 1.93$ and $2.01(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{Ac}), 3.82-3.99(3$ $\mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}$ and $\left.6^{\prime}-\mathrm{H}_{2}\right), 4.32\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.5,4-\mathrm{H}\right), 4.44(1 \mathrm{H}$, dd, $\left.J_{5,6 \mathrm{a}} 4.5, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.5,6-\mathrm{H}^{\mathrm{a}}\right), 4.52$ and $4.81(2 \mathrm{H}, \mathrm{AB} \mathrm{q}, J 11.6$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 4.58-4.70\left(3 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}^{\mathrm{b}}\right), 4.91(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2^{\prime}, 3^{\prime}} 10.5,3^{\prime}-\mathrm{H}\right), 5.02(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 1-\mathrm{H}), 5.15\left(1 \mathrm{H}\right.$, dd, $J_{1^{\prime}, 2^{\prime}} 7.5$, $\left.2^{\prime}-\mathrm{H}\right), 5.25\left(1 \mathrm{H}, \mathrm{d}, J_{3^{\prime}, 4^{\prime}} 3.5,4^{\prime}-\mathrm{H}\right), 5.46(1 \mathrm{H}, \mathrm{d}, 2-\mathrm{H}), 5.66(1 \mathrm{H}$, $\left.\mathrm{t}, J_{2,3}=J_{3,4}=3.3,3-\mathrm{H}\right)$ and $7.15-8.15(20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph})$.

2,3,4,6-Tetra-O-acetyl- $\beta$-d-galactopyranosyl-( $\mathbf{1} \rightarrow \mathbf{4}$ )-2,3,6-tri- $O$ -benzoyl- $\alpha$-D-altropyranosyl hydrogenphosphonate, triethylammonium salt 23
A solution of the disaccharide $22(0.455 \mathrm{~g}, 0.499 \mathrm{mmol})$ in THF $\left(5 \mathrm{~cm}^{3}\right)$ containing $20 \% \mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}(200 \mathrm{mg})$ was shaken under a slight overpressure of $\mathrm{H}_{2}$ at rt for 2 h . The catalyst was filtered off through a Celite pad and the filtrate was concentrated to give the $\alpha, \beta$-hemiacetal $24(0.39 \mathrm{~g}, 95 \%)$ as an amorphous solid [ $R_{\mathrm{f}} 0.3$ (solvent $C$ ); $\delta_{\mathrm{H}}(500 \mathrm{MHz}) 1.89-2.03(12 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ac})$, $3.78-3.92\left(3 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}\right.$ and $\left.6^{\prime}-\mathrm{H}_{2}\right), 4.30\left(\mathrm{dd}, J_{4,5} 8.5,4-\mathrm{H}^{\beta}\right)$,
$4.39\left(\mathrm{dd}, J_{4,5} 8.5,4-\mathrm{H}^{\alpha}\right), 4.42\left(\mathrm{dt}, J_{5,6 \mathrm{a}}=J_{5,6 \mathrm{~b}}=3.0,5-\mathrm{H}^{\mathrm{\beta}}\right), 4.52$ (dd, $\left.J_{5,6 \mathrm{a}} 3.0, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.7,6-\mathrm{H}^{\mathrm{a}}\right), 4.61\left(\mathrm{~d}, J_{1^{\prime}, 2^{\prime}} 7.0,1^{\prime}-\mathrm{H}^{\mathrm{c}}\right), 4.67(\mathrm{~d}$, $\left.J_{1^{\prime}, 2^{\prime}} 7.0,1^{\prime}-\mathrm{H}^{\beta}\right), 4.70-4.78\left(\mathrm{~m}, 5-\mathrm{H}^{\mathrm{a}}\right.$ and $\left.6-\mathrm{H}^{\mathrm{b}}\right)$, $4.92\left(\mathrm{dd}, J_{3^{\prime}, 4^{\prime}}\right.$ $3.0,3^{\prime}-\mathrm{H}^{\alpha}$ ), 4.95 (dd, $J_{3^{\prime}, 4} 3.0,3^{\prime}-\mathrm{H}^{\beta}$ ), $5.13\left(\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 9.0,2^{\prime}-\mathrm{H}^{\alpha}\right)$, $5.17\left(\mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 9.0,2^{\prime}-\mathrm{H}^{\beta}\right), 5.22\left(\mathrm{~d}, 4^{\prime}-\mathrm{H}^{\alpha}\right), 5.26\left(\mathrm{~d}, 4^{\prime}-\mathrm{H}^{\beta}\right), 5.30$ (br s, $1-\mathrm{H}^{\beta}$ ), $5.39\left(\mathrm{br} \mathrm{s}, 1-\mathrm{H}^{\alpha}\right), 5.46\left(\mathrm{~d}, 2-\mathrm{H}^{\beta}\right), 5.51\left(\mathrm{~d}, 2-\mathrm{H}^{\alpha}\right), 5.69$ $\left(\mathrm{t}, J_{2,3}=J_{3,4}=2.8,3-\mathrm{H}^{\beta}\right), 5.79\left(\mathrm{t}, J_{2,3}=J_{3,4}=2.8,3-\mathrm{H}^{\omega}\right)$ and $7.20-$ $8.20(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph}) ; \alpha: \beta=0.8: 1]$.
The reaction of the compound $24(0.39 \mathrm{~g}, 0.474 \mathrm{mmol})$ with $\mathrm{PCl}_{3}\left(0.165 \mathrm{~cm}^{3}, 1.89 \mathrm{mmol}\right)$, imidazole ( $0.45 \mathrm{~g}, 6.62 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}\left(0.99 \mathrm{~cm}^{3}, 7.09 \mathrm{mmol}\right)$ in $\mathrm{CH}_{3} \mathrm{CN}\left(10 \mathrm{~cm}^{3}\right)$, followed by hydrolysis with $1 \mathrm{~mol} \mathrm{dm}^{-3}$ aq. TEA hydrogen carbonate ( 2.5 $\mathrm{cm}^{3}$ ), was accomplished as described for the preparation of the disaccharide H-phosphonate 13. After work-up, the solution was concentrated and acetonitrile was evaporated off from the residue. The residue was dissolved in the same solvent $\left(5 \mathrm{~cm}^{3}\right)$ and anhydrous $\mathrm{H}_{3} \mathrm{PO}_{3}(0.39 \mathrm{~g}, 4.73 \mathrm{mmol})$ was added to the solution. The reaction mixture was kept at rt for 20 h , then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(50 \mathrm{~cm}^{3}\right)$ and washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and $0.5 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ aq. TEA hydrogen carbonate, dried by filtration through cotton wool, and concentrated. $\mathrm{FCC}\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH},(99: 1) \longrightarrow(80: 20)\right]$ gave the H-phosphonate $23(0.175 \mathrm{~g}, 35 \%$ from the disaccharide 22) as an amorphous solid, $[a]_{\mathrm{D}}^{25}+1\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.35$ (solvent $F$ ); $\delta_{\mathrm{H}}$ $(200 \mathrm{MHz}) 1.20\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.91,1.92,1.93$ and 2.02 $(12 \mathrm{H}, 4 \times \mathrm{s}, 4 \times \mathrm{Ac}), 2.91\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.80-3.88(2 \mathrm{H}$, $\mathrm{m}, 5^{\prime}-\mathrm{H}$ and $\left.6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 3.93\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 66^{\prime}} 7.8, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 13.5,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right)$, $4.35\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.6,4-\mathrm{H}\right), 4.46\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.6,6-\mathrm{H}^{\mathrm{a}}\right), 4.64$ $\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.7,1^{\prime}-\mathrm{H}\right), 4.71\left(1 \mathrm{H}, \mathrm{dd}, J_{5,66} 1.0,6-\mathrm{H}^{\mathrm{b}}\right), 4.88(1 \mathrm{H}$, ddd, $\left.J_{5,6 \mathrm{a}} 3.7,5-\mathrm{H}\right), 4.91\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4} 3.2,3^{\prime}-\mathrm{H}\right), 5.13(1 \mathrm{H}$, dd, $\left.J_{2^{\prime}, 3^{\prime}} 10.6,2^{\prime}-\mathrm{H}\right), 5.23\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right)$, $5.44(1 \mathrm{H}, \mathrm{d}, 2-\mathrm{H}), 5.67(1$ $\left.\mathrm{H}, \mathrm{t}, J_{2,3}=J_{3,4}=2.8,3-\mathrm{H}\right), 5.72\left(1 \mathrm{H}, \mathrm{d}, J_{1, \mathrm{p}} 8.5,1-\mathrm{H}\right), 7.02(1 \mathrm{H}$, d, $\left.J_{\mathrm{H}, \mathrm{P}} 640.0, \mathrm{HP}\right)$ and $7.40-8.20(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph})$; $\delta_{\mathrm{P}} 0.58$; ESMS(-): $m / z 884.9\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$(expected $m / z$, 885.008. $\mathrm{C}_{47} \mathrm{H}_{58} \mathrm{NO}_{20} \mathrm{P}$ requires $M, 987.208$ ). Also isolated was the disaccharide hemiacetal $24(0.2 \mathrm{~g}, 49 \%$ recovery $)$.

Dec-9-enyl 2,3,4,6-tetra-O-acetyl- $\beta$-d-galactopyranosyl-(1 $\rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-D-altropyranosyl phosphate, triethylammonium salt 26

This compound was prepared by condensation of the glycobiosyl H-phosphonate 23 ( $98 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) and dec-9-en-1-ol ( $0.053 \mathrm{~cm}^{3}, 0.30 \mathrm{mmol}$ ) in pyridine $\left(1 \mathrm{~cm}^{3}\right)$ in the presence of trimethylacetyl chloride ( $0.037 \mathrm{~cm}^{3}, 0.30 \mathrm{mmol}$ ), followed by oxidation with iodine ( $51 \mathrm{mg}, 0.20 \mathrm{mmol}$ ) in pyridine-water ( $95: 5 ; 2 \mathrm{~cm}^{3}$ ) as described for the synthesis of the phosphodiester 15. $\mathrm{FCC}\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH},(99: 1) \longrightarrow(80: 20)\right]$ gave the phosphodiester 26 ( $85 \mathrm{mg}, 75 \%$ ) as an amorphous solid, $[\alpha]_{\mathrm{D}}^{25}+2$ ( c 1, $\mathrm{CHCl}_{3}$ ); $R_{\mathrm{f}} 0.5$ (solvent $F$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.25(19 \mathrm{H}, \mathrm{m}$, $3 \times \mathrm{MeCH}_{2}$ and $\left.5 \times \mathrm{CH}_{2}\right), 1.48\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)$, $1.95(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Ac}), 1.97$ and $2.00(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{Ac}), 2.03$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 2.95\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.74-3.96$ $\left(5 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}, 6^{\prime}-\mathrm{H}_{2}\right.$ and $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}\right)$, $4.38\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.6\right.$, $4-\mathrm{H}), 4.46\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{a}} 3.0, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.8,6-\mathrm{H}^{\mathrm{a}}\right), 4.64\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}}\right.$ $\left.7.8,1^{\prime}-\mathrm{H}\right), 4.72\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.0,6-\mathrm{H}^{\mathrm{b}}\right), 4.87-4.95\left(3 \mathrm{H}, \mathrm{m}, 3^{\prime}-\right.$, $5-\mathrm{H}$ and $H \mathrm{CH}=\mathrm{CH}), 4.98\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.6,{ }^{3} J_{\mathrm{H}, \mathrm{H}-E} 16.8\right.$, $\mathrm{HCH}=\mathrm{CH})$, $5.12\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.1,2^{\prime}-\mathrm{H}\right), 5.23\left(1 \mathrm{H}, \mathrm{d}, J_{3^{\prime}, 4^{\prime}}\right.$ $\left.3.1,4^{\prime}-\mathrm{H}\right), 5.50\left(1 \mathrm{H}, \mathrm{d}, J_{2,3} 3.0,2-\mathrm{H}\right), 5.64\left(1 \mathrm{H}, \mathrm{d}, J_{1, \mathrm{P}} 7.7,1-\mathrm{H}\right)$, $5.68\left(1 \mathrm{H}, \mathrm{dd}, J_{3,4} 3.5,3-\mathrm{H}\right), 5.80\left(1 \mathrm{H}, \mathrm{ddt}, J_{\mathrm{H}, \mathrm{CH}}, 6.6,{ }^{3} J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}}\right.$ $\left.10.9, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$ and $7.40-8.25(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph}) ; \delta_{\mathrm{P}}-2.79$; ESMS(-): $m / z 1039.0\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$(expected $m / z$, 1039.14. $\mathrm{C}_{57} \mathrm{H}_{76} \mathrm{NO}_{21} \mathrm{P}$ requires $M, 1141.344$ ).

## Dec-9-enyl $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-d-altropyranosyl phosphate, triethylammonium salt 7

To a solution of compound $\mathbf{2 6}(50 \mathrm{mg})$ in $\mathrm{MeOH}\left(1.8 \mathrm{~cm}^{3}\right)$ was added $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ methanolic $\mathrm{NaOMe}\left(0.2 \mathrm{~cm}^{3}\right)$. The mixture (now $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ in NaOMe ) was kept at $0{ }^{\circ} \mathrm{C}$ for 16 h and then at room temperature for 8 h , whereafter it was deionized
with Dowex $50 \mathrm{~W}-\mathrm{X} 4\left(\mathrm{H}^{+}\right)$resin, filtered and immediately neutralized with $\mathrm{Et}_{3} \mathrm{~N}$. After concentration, water ( $3 \times 5 \mathrm{~cm}^{3}$ ) was evaporated off from the residue to remove methyl benzoate. The phosphodiester $7(28 \mathrm{mg}, 96 \%)$ was thereby obtained as an amorphous solid, $[a]_{\mathrm{D}}^{25}+36(c 1, \mathrm{MeOH}) ; R_{\mathrm{f}} 0.65$ (solvent $G$ ); $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) $1.15\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.23$ $\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.52\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.94$ ( $\left.2 \mathrm{H}, \mathrm{dt}, J 6.7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 3.10\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right)$, $4.41\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.0,1^{\prime}-\mathrm{H}\right), 5.21\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J_{1, \mathrm{P}} 6.6,1-\mathrm{H}\right)$ and $5.83\left(1 \mathrm{H}\right.$, ddt, $J_{\mathrm{H}, \mathrm{CH}} 6.7, J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}} 10.1, J_{\mathrm{H}, \mathrm{H}-E} 18.0, \mathrm{CH}_{2}-$ $\left.\mathrm{C} H=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}, \delta_{\mathrm{P}}$ and ESMS(-) data: see Table 1.

Methyl 2,3,4,6-tetra- $O$-acetyl- $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )-2,3-$O$-isopropylidene- $\alpha$-D-rhamnopyranoside 28
A solution of acetobromogalactose $21(1.49 \mathrm{~g}, 3.58 \mathrm{mmol})$ in acetonitrile-toluene ( $10: 3 ; 13 \mathrm{~cm}^{3}$ ) was added to a stirred mixture of methyl $2,3-O$-isopropylidene- $\alpha$-d-rhamnopyranoside ${ }^{13}$ $27(0.318 \mathrm{~g}, 1.45 \mathrm{mmol}), \mathrm{Hg}(\mathrm{CN})_{2}(0.9 \mathrm{~g}, 3.58 \mathrm{mmol})$ and $\mathrm{HgBr}_{2}$ ( $0.64 \mathrm{~g}, 1.79 \mathrm{mmol}$ ) in the same mixed solvent ( $5 \mathrm{~cm}^{3}$ ). After being stirred at rt for 16 h , the reaction mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(50 \mathrm{~cm}^{3}\right)$, washed successively with $1 \mathrm{~mol} \mathrm{dm}^{-3}$ aq. KBr , saturated aq. $\mathrm{NaHCO}_{3}$, and water, dried by filtration through cotton wool, and concentrated. FCC [toluene-ethyl acetate, (8:2)] of the residue gave the disaccharide derivative 28 ( 0.59 g , $74 \%$ ), mp 126-129 ${ }^{\circ} \mathrm{C}$ (from ethanol); $[a]_{\mathrm{D}}^{22}+25.7$ ( c $1, \mathrm{CHCl}_{3}$ ) (Found: C, 52.5; H, 6.6. $\mathrm{C}_{24} \mathrm{H}_{36} \mathrm{O}_{14}$ requires C, $52.6 ; \mathrm{H}, 6.6 \%$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.23\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.3,6-\mathrm{H}_{3}\right), 1.32$ and $1.50(6 \mathrm{H}$, $2 \times \mathrm{s}, \mathrm{CMe}_{2}$ ), 2.00, 2.05, 2.07 and $2.15(12 \mathrm{H}, 4 \times \mathrm{s}, 4 \times \mathrm{Ac})$, $3.34(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.36\left(1 \mathrm{H}, \mathrm{dd}, J_{3,4} 7.2,4-\mathrm{H}\right), 3.63(1 \mathrm{H}, \mathrm{dq}$, $\left.J_{4,5} 9.8,5-\mathrm{H}\right), 3.88\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6^{\prime}} 6.6,5^{\prime}-\mathrm{H}\right), 4.06\left(1 \mathrm{H}, \mathrm{d}, J_{2,3} 5.7\right.$, $2-\mathrm{H}), 4.14\left(2 \mathrm{H}, \mathrm{d}, 6^{\prime}-\mathrm{H}_{2}\right), 4.24(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 4.65\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}}\right.$ $\left.8.0,1^{\prime}-\mathrm{H}\right), 4.81(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H}), 5.00\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.2,3^{\prime}-\mathrm{H}\right), 5.21$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.3,2^{\prime}-\mathrm{H}\right)$ and $5.36\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right)$.

## 2,3,4,6-Tetra- $O$-acetyl $\beta$ - -d-galactopyranosyl-( $1 \rightarrow 4$ )-1,2,3-tri-$O$-acetyl- $\alpha$-d-rhamnopyranose 29

To a stirred solution of the disaccharide $28(0.728 \mathrm{~g})$ in chloroform ( $36 \mathrm{~cm}^{3}$ ) was added $90 \%$ aq. trifluoroacetic acid $\left(4 \mathrm{~cm}^{3}\right)$. Stirring was continued for 2 h , whereafter the solution was concentrated and toluene was twice evaporated off from the residue. The residue was then dissolved in $\mathrm{Ac}_{2} \mathrm{O}\left(5 \mathrm{~cm}^{3}\right)$ and a sulfuric acid-acetic anhydride mixture [1:50(v/v); $10.2 \mathrm{~cm}^{3}$ ] was added. The solution was stirred at rt for 2 h , whereafter it was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(100 \mathrm{~cm}^{3}\right)$, washed successively with water, saturated aq. $\mathrm{NaHCO}_{3}$, and water, dried by filtration through cotton wool, and concentrated. Toluene was twice evaporated off from the residue. FCC (solvent $B \longrightarrow$ solvent C) of the residue gave the heptaacetate $29(0.57 \mathrm{~g}, 69 \%)$, mp $145-148{ }^{\circ} \mathrm{C}$ (from ethanol); $[a]_{\mathrm{D}}^{21}+36.8\left(c 0.99, \mathrm{CHCl}_{3}\right)$ (Found: C, 50.6 ; H, 5.9. $\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{O}_{17}$ requires C, $50.3 ; \mathrm{H}, 5.9 \%$ ); $\delta_{\mathrm{H}}(200$ $\mathrm{MHz}) 1.30\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.1,6-\mathrm{H}_{3}\right), 1.96,2.02,2.03,2.04$ and 2.14 $(15 \mathrm{H}, 5 \times \mathrm{s}, 5 \times \mathrm{Ac}), 2.13(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Ac}), 3.63(1 \mathrm{H}, \mathrm{t}$, $\left.J_{3,4}=J_{4,5}=9.4,4-\mathrm{H}\right), 3.82(1 \mathrm{H}, \mathrm{dq}, 5-\mathrm{H}), 3.87\left(1 \mathrm{H}\right.$, ddd, $J_{5^{\prime}, 6 a^{\prime}}$ $\left.7.1,5^{\prime}-\mathrm{H}\right), 4.02\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 11.0,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.16\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{~b}}\right.$ $\left.6.4,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 4.58\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.8,1^{\prime}-\mathrm{H}\right), 4.97\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.4\right.$, $\left.3^{\prime}-\mathrm{H}\right), 5.14\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.3,2^{\prime}-\mathrm{H}\right), 5.19\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.3\right.$, $2-\mathrm{H}), 5.28(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 5.33\left(1 \mathrm{H}, \mathrm{dd}, J_{4^{\prime}, 5^{\prime}} 0.5,4^{\prime}-\mathrm{H}\right)$ and 5.94 $\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.9,1-\mathrm{H}\right)$.

## 2,3,4,6-Tetra-O-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow \mathbf{4}$ )-2,3-di- $O$ -acetyl- $\alpha$-D-rhamnopyranose 30

This compound was prepared from compound $29(0.307 \mathrm{~g})$ as described for the hemiacetal derivative 12. Two consecutive FCC separations [toluene-ethyl acetate, $(55: 45) \longrightarrow(8: 2)$ and solvent $B \longrightarrow$ solvent $D$ ] gave the disaccharide $\alpha$-hemiacetal $\mathbf{3 0}$ $(0.19 \mathrm{~g}, 66 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{21}+11\left(c 0.97, \mathrm{CHCl}_{3}\right)$ (Found: C, 49.4; H, 6.0. $\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{16}$ requires C, 49.8; H, 5.9\%); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.28\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.2,6-\mathrm{H}_{3}\right), 1.96,2.01,2.03,2.04$,
2.12 and $2.14(18 \mathrm{H}, 6 \times \mathrm{s}, 6 \times \mathrm{Ac}), 3.59\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.4\right.$, $4-\mathrm{H}), 3.87\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}}=J_{5^{\prime}, 6 \mathrm{~b}^{\prime}}=6.5,5^{\prime}-\mathrm{H}\right), 4.00(1 \mathrm{H}, \mathrm{dq}, 5-\mathrm{H})$, $4.01\left(1 \mathrm{H}, \mathrm{dd}, 6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.16\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 11.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 4.57$ $\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2}, 7.8,1^{\prime}-\mathrm{H}\right), 4.97\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4} 3.3,3^{\prime}-\mathrm{H}\right), 5.08(1 \mathrm{H}$, d, $\left.J_{1,2} 1.9,1-\mathrm{H}\right), 5.13\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.5,2^{\prime}-\mathrm{H}\right), 5.21(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2,3} 3.5,2-\mathrm{H}\right), 5.32\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right)$ and $5.33(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H})$.

##  acetyl- $\alpha$-D-rhamnopyranosyl hydrogenphosphonate, triethylammonium salt 31

This compound was prepared by the reaction of the hemiacetal $30(0.184 \mathrm{~g}, 0.32 \mathrm{mmol})$ with $\mathrm{PCl}_{3}\left(0.166 \mathrm{~cm}^{3}, 1.91 \mathrm{mmol}\right)$, imidazole ( $0.433 \mathrm{~g}, 6.36 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}\left(0.93 \mathrm{~cm}^{3}, 6.7 \mathrm{mmol}\right)$ in acetonitrile $\left(15 \mathrm{~cm}^{3}\right)$ followed by hydrolysis as described for the H-phosphonate 13. For work-up, the reaction mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(100 \mathrm{~cm}^{3}\right)$ and washed successively with cold saturated aq. $\mathrm{NaHCO}_{3}\left(2 \times 40 \mathrm{~cm}^{3}\right)$ and cold $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ aq. TEA hydrogen carbonate $\left(2 \times 40 \mathrm{~cm}^{3}\right)$. The organic phase was discarded. The aqueous washings were combined, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(4 \times 20 \mathrm{~cm}^{3}\right)$. The combined organic washings were dried by filtration through cotton wool, and concentrated to produce the H-phosphonate $31(0.165 \mathrm{~g}, 70 \%)$ as a chromatographically homogeneous amorphous solid, $[a]_{D}^{25}$ $+23.1\left(c 1.06, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.27\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.3\right.$, $\left.6-\mathrm{H}_{3}\right), 1.30\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.94,1.95,2.00,2.02,2.08$ and $2.11(18 \mathrm{H}, 6 \times \mathrm{s}, 6 \times \mathrm{Ac}), 3.03\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.56(1 \mathrm{H}$, $\left.\mathrm{t}, J_{3,4}=J_{4,5}=9.4,4-\mathrm{H}\right), 3.83\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}}=J_{5^{\prime}, 6 \mathrm{~b}^{\prime}}=6.6,5^{\prime}-\mathrm{H}\right)$, $3.98(1 \mathrm{H}, \mathrm{dq}, 5-\mathrm{H}), 3.99\left(1 \mathrm{H}, \mathrm{dd}, 6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.12\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}}\right.$ $\left.11.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right)$, $4.54\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.7,1^{\prime}-\mathrm{H}\right), 4.94\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.3\right.$, $\left.3^{\prime}-\mathrm{H}\right), 5.10\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.5,2^{\prime}-\mathrm{H}\right), 5.21\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.8\right.$, $2-\mathrm{H}), 5.30\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right), 5.31\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.6,3-\mathrm{H}\right), 5.44(1 \mathrm{H}$, dd, $\left.J_{1, \mathrm{P}} 8.8,1-\mathrm{H}\right)$ and $5.44\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{H}, \mathrm{P}} 642.0, \mathrm{HP}\right) ; \delta_{\mathrm{P}}-0.10$; ESMS(-): $m / z 641.0\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$(expected $m / z$, 641.07. $\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{NO}_{18} \mathrm{P}$ requires $M, 743.27$ ).

Dec-9-enyl 2,3,4,6-tetra-O-acetyl- $\beta$-d-galactopyranosyl-( $1 \rightarrow 4$ )-2,3-di- O-acetyl- $\alpha$-d-rhamnopyranosyl phosphate, triethylammonium salt 32

This compound was prepared by condensation of the H-phosphonate $31(0.14 \mathrm{~g}, 0.19 \mathrm{mmol})$ and dec-9-en-1-ol $\left(0.067 \mathrm{~cm}^{3}\right.$, 0.38 mmol ) in pyridine $\left(1 \mathrm{~cm}^{3}\right)$ in the presence of trimethylacetyl chloride ( $0.058 \mathrm{~cm}^{3}, 0.47 \mathrm{mmol}$ ) followed by oxidation with iodine ( $0.096 \mathrm{~g}, 0.376 \mathrm{mmol}$ ) in pyridine-water $\left(95: 5 ; 2 \mathrm{~cm}^{3}\right)$ as described for the synthesis of the phosphodiester 15. FCC $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}, \quad(99: 0: 1) \longrightarrow(89: 10: 1)\right]$ gave the phosphodiester $32(0.148 \mathrm{~g}, 88 \%)$ as an amorphous solid, $[a]_{D}^{27}$ $+12.4\left(c 1.09, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.18\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.0\right.$, $\left.6-\mathrm{H}_{3}\right), 1.24\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.25\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.53$ $\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.89,1.90,1.97,1.98,2.03$ and $2.07(18 \mathrm{H}, 6 \times \mathrm{s}, 6 \times \mathrm{Ac}), 1.96\left(2 \mathrm{H}, \mathrm{dt}, J 6.8, \mathrm{CH}_{2} \mathrm{CH} \mathrm{H}_{2} \mathrm{CH}=\right)$, $3.00\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.51\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 9.7,4-\mathrm{H}\right), 3.71-3.86$ $\left(4 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}, 6^{\prime}-\mathrm{H}^{\mathrm{a}}\right.$ and $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.97(1 \mathrm{H}, \mathrm{dq}, 5-\mathrm{H}), 4.08$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 6.6, J_{6 \mathrm{a}^{\prime} ; 6 \mathrm{~b}^{\prime}} 11.2,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 4.52\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.7\right.$, $\left.1^{\prime}-\mathrm{H}\right), 4.84\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.6,{ }^{3} J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}} 9.3, H \mathrm{CH}=\mathrm{CH}\right), 4.90$ $\left(1 \mathrm{H}, \mathrm{dd},{ }^{3} J_{\mathrm{H}, \mathrm{H}-\mathrm{E}} 17.0, \mathrm{HCH}=\mathrm{CH}\right), 4.92\left(1 \mathrm{H}, \mathrm{dd}, 3^{\prime}-\mathrm{H}\right), 5.08(1 \mathrm{H}$, dd, $\left.J_{2^{\prime}, 3^{\prime}} 10.5,2^{\prime}-\mathrm{H}\right), 5.19\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.3,2-\mathrm{H}\right), 5.25(1 \mathrm{H}, \mathrm{d}$, $\left.J_{3^{\prime}, 4^{\prime}} 4.0,4^{\prime}-\mathrm{H}\right), 5.26\left(1 \mathrm{H}, \mathrm{dd}, J_{3,4} 9.4,3-\mathrm{H}\right), 5.34\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2}\right.$ $1.5, J_{1, \mathrm{P}} 8.8,1-\mathrm{H}$ ) and $5.73\left(1 \mathrm{H}\right.$, ddt, $\left.J_{\mathrm{H}, \mathrm{CH}} 6.8, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$; $\delta_{\mathrm{P}}-3.12 ; \operatorname{ESMS}(-): m / z 795.3\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$ (expected $m / z, 795.21 . \mathrm{C}_{40} \mathrm{H}_{68} \mathrm{NO}_{19} \mathrm{P}$ requires $M, 897.41$ ).

## Dec-9-enyl $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-rhamnopyranosyl phosphate, ammonium salt 8

De- $O$-acetylation of compound $\mathbf{3 2}(74 \mathrm{mg})$ with $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ NaOMe in methanol ( 3 h at rt ) followed by work-up, as described in the preparation of the phosphodiester $\mathbf{6}$, produced a crude product, which then was applied to a column ( $18 \times 1.5$ $\mathrm{cm})$ of Fractogel TSK DEAE-650 (S) ( $\mathrm{HCO}_{3}{ }^{-}$-form) (Merck)
eluted with a linear gradient of $\mathrm{NH}_{4} \mathrm{HCO}_{3}(0 \longrightarrow 0.1 \mathrm{~mol}$ $\mathrm{dm}^{-3}$ ) in 3:2 water-propan-2-ol at $1 \mathrm{~cm}^{3} \mathrm{~min}^{-1}$ to afford the phosphodiester 8 ( $41 \mathrm{mg}, 88 \%$ ) as an amorphous solid, $[a]_{\mathrm{D}}^{26}$ $+22.4(c 0.99, \mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) $1.22-$ $1.45\left(13 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{3}\right.$ and $\left.5 \times \mathrm{CH}_{2}\right), 1.63\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.05\left(2 \mathrm{H}, \mathrm{dt}, J 6.9, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 4.48\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}}\right.$ $\left.7.0,1^{\prime}-\mathrm{H}\right), 5.32\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J_{1, \mathrm{P}} 6.3,1-\mathrm{H}\right)$ and $5.91(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}, \delta_{\mathrm{P}}$ and $\operatorname{ESMS}(-)$ data: see Table 1.

## $\mathbf{2 , 3 , 4}$-Tri- $\boldsymbol{O}$-benzoyl- $\boldsymbol{\alpha}, \boldsymbol{\beta}$-d-fucopyranose $\mathbf{3 4}$

To a solution of 1,2,3,4-tetra-O-benzoyl- $\alpha$-D-fucopyranose 33 ( $0.5 \mathrm{~g}, 0.861 \mathrm{mmol}$ ) [prepared by standard benzoylation of $\alpha-\mathrm{D}-$ fucose with benzoyl chloride in pyridine-chloroform; $\delta_{\mathrm{H}}$ (200 $\mathrm{MHz}) 1.34\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.4,6-\mathrm{H}_{3}\right), 4.67(1 \mathrm{H}, \mathrm{q}, 5-\mathrm{H}), 5.92(1 \mathrm{H}$, d, $4-\mathrm{H}), 6.01\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 10.8,2-\mathrm{H}\right), 6.14\left(1 \mathrm{H}, \mathrm{dd}, J_{3,4} 3.0\right.$, $3-\mathrm{H}), 6.91\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 3.3,1-\mathrm{H}\right)$ and $7.17-8.28(20 \mathrm{H}, \mathrm{m}$, $4 \times \mathrm{Ph})]$ in acetonitrile $\left(3 \mathrm{~cm}^{3}\right)$ was added $2 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{Me}_{2} \mathrm{NH}$ in THF ( $\left.4.3 \mathrm{~cm}^{3} ; 8.61 \mathrm{mmol}\right)$ and the mixture was kept at rt with monitoring by TLC (solvents $B$ and $C$ ). After $16-24 \mathrm{~h}$, the mixture was concentrated to dryness and acetonitrile was evaporated off from the residue. FCC [toluene-ethyl acetate, $(99: 1) \longrightarrow(85: 15)]$ gave the unchanged starting material 33 ( $0.079 \mathrm{~g}, 16 \%$ recovery) and the hemiacetal $34(0.25 \mathrm{~g}, 61 \%$; amorphous solid), $[a]_{\mathrm{D}}^{22}+247.4\left(c 1, \mathrm{CHCl}_{3}\right.$ ) (Found: C, $68.3 ; \mathrm{H}$, 5.1. $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{8}$ requires $\mathrm{C}, 68.1 ; \mathrm{H}, 5.1 \%$ ); $\delta_{\mathrm{H}}(200 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}+\mathrm{D}_{2} \mathrm{O}\right) 1.26\left(\mathrm{~d}, J_{5,6} 6.4,6-\mathrm{H}^{\mathrm{u}}\right), 1.35\left(\mathrm{~d}, J_{5,6} 6.3,6-\mathrm{H}^{\beta}\right)$, $4.12\left(\mathrm{dq}, J_{4,5} 0.8,5-\mathrm{H}^{\beta}\right), 4.67\left(\mathrm{q}, \mathrm{H}-5^{\alpha}\right), 5.06\left(\mathrm{~d}, J_{1,2} 6.8,1-\mathrm{H}^{\beta}\right)$, $5.66-5.85\left(\mathrm{~m}, 1-\mathrm{H}^{\alpha}, 2-\mathrm{H}, 3-\mathrm{H}^{\beta}\right.$ and $\left.4-\mathrm{H}\right), 6.09\left(\mathrm{dd}, J_{2,3} 10.6, J_{3,4}\right.$ $\left.3.2,3-\mathrm{H}^{\alpha}\right)$ and $7.01-8.40(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph}) ; \alpha: \beta=5: 1$.

## 2,3,4-Tri- $O$-benzoyl- $\alpha$-d-fucopyranosyl trichloroacetimidate $\mathbf{3 5}$

To a stirred solution of the hemiacetal $34(0.228 \mathrm{~g}, 0.48 \mathrm{mmol})$ and $\mathrm{CCl}_{3} \mathrm{CN}\left(2 \mathrm{~cm}^{3}, 20 \mathrm{mmol}\right)$ in dichloromethane ( $4 \mathrm{~cm}^{3}$ ) cooled to $0^{\circ} \mathrm{C}$ was added $\operatorname{DBU}\left(0.072 \mathrm{~cm}^{3}, 0.48 \mathrm{mmol}\right)$ under argon. The mixture was stirred for 2 h at $0^{\circ} \mathrm{C}$ and then concentrated. FCC (solvent $A$ ) of the residue gave the $\alpha$-fucopyranosyl trichloroacetimidate $35(0.277 \mathrm{~g}, 93 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{22}+209.4\left(c 1, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.34\left(3 \mathrm{H}, \mathrm{d}, J_{5,6} 6.5\right.$, $\left.6-\mathrm{H}_{3}\right), 4.66(1 \mathrm{H}, \mathrm{dq}, 5-\mathrm{H}), 5.89\left(1 \mathrm{H}, \mathrm{dd}, J_{4,5} 0.5,4-\mathrm{H}\right), 5.92$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 10.5,2-\mathrm{H}\right), 6.06\left(1 \mathrm{H}, \mathrm{dd}, J_{3,4} 3.2,3-\mathrm{H}\right), 6.86(1 \mathrm{H}$, d, $\left.J_{1,2} 3.3,1-\mathrm{H}\right), 7.10-8.23(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph})$ and $8.60(1 \mathrm{H}, \mathrm{s}$, $\mathrm{HN}) ; \delta_{\mathrm{C}} 16.19$ (C-6), 67.95 (C-5), 68.05 (C-2), 68.66 (C-3), 71.29 (C-4), $90.91\left(\mathrm{CCl}_{3}\right), 94.11(\mathrm{C}-1), 128.30-133.48(\mathrm{Ph}), 160.78$ $(\mathrm{C}=\mathrm{NH})$ and 165.62-165.83 (C=O); ESMS(+): m/z 459.2 $\left(100 \%\right.$, $\left[\mathrm{M}-\mathrm{CCl}_{3} \mathrm{CONH}\right]^{+}$) (expected $\mathrm{m} / \mathrm{z}$, 459.47. $\mathrm{C}_{29} \mathrm{H}_{24}{ }^{-}$ $\mathrm{Cl}_{3} \mathrm{NO}_{8}$ requires $M, 620.86$ ).

## 2,3,4-Tri- $O$-benzoyl- $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )-1,2,3,6-tetra- $O$ -benzoyl- $\alpha$-d-mannopyranose 37 and 2,3,4-tri- $O$-benzoyl- $\alpha$-d-fucopyranosyl-( $1 \rightarrow 4$ )-1,2,3,6-tetra- $O$-benzoyl- $\alpha$-D-mannopyranose 40

A mixture of the trichloroacetimidate $35(0.554 \mathrm{~g}, 0.89 \mathrm{mmol})$, 1,2,3,6-tetra- $O$-benzoyl- $\alpha$-d-mannopyranose ${ }^{3} 36$ ( $0.638 \mathrm{~g}, 1.07$ mmol ) and freshly activated molecular sieves $4 \AA$ (powder, 1 g ) in dry dichloromethane $\left(5 \mathrm{~cm}^{3}\right.$ ) was stirred under argon for 30 min . TMS triflate $\left(0.046 \mathrm{~cm}^{3}, 0.22 \mathrm{mmol}\right)$ was then added and the mixture was cooled to $-70^{\circ} \mathrm{C}$. Stirring was continued for a further 1.5 h , while the mixture slowly warmed to $-10^{\circ} \mathrm{C}$. The reaction was quenched with a few drops of $N, N-$ diisopropylethylamine, the solids were filtered off, and the solvent was removed under reduced pressure. FCC (toluene $\longrightarrow$ solvent $B$ ) of the residue gave a mixture of the disaccharides 37 and 40, which were then separated by further FCC [dichloromethane-ethyl acetate, $(100: 0) \longrightarrow(98: 2)$ ]. That provided, first, the $\beta$-linked disaccharide $37(0.582 \mathrm{~g}, 62 \%)$ as an amorphous solid, $[a]_{D}^{23}+118\left(c 1.03, \mathrm{CHCl}_{3}\right)$ (Found: C, 69.6; H, 4.7. $\mathrm{C}_{61} \mathrm{H}_{50} \mathrm{O}_{17}$ requires C, $\left.69.4 ; \mathrm{H}, 4.8 \%\right)$; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 0.82$ $\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.3,6^{\prime}-\mathrm{H}_{3}\right), 3.56\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.27\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}}\right.$
2.6, $5-\mathrm{H}), 4.45\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.7,6-\mathrm{H}^{\mathrm{a}}\right), 4.68\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{bb}} 1.8\right.$, $\left.6-\mathrm{H}^{\mathrm{b}}\right), 4.69\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.7,4-\mathrm{H}\right), 4.96\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2}, 7.9\right.$, $\left.1^{\prime}-\mathrm{H}\right), 5.39\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.3,3^{\prime}-\mathrm{H}\right), 5.48\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right), 5.70$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.3,2^{\prime}-\mathrm{H}\right), 5.85\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.3,2-\mathrm{H}\right), 5.99(1 \mathrm{H}$, dd, $3-\mathrm{H})$, $6.48\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.7,1-\mathrm{H}\right)$ and $7.08-8.22(35 \mathrm{H}, \mathrm{m}$, $7 \times \mathrm{Ph}$ ). Continued elution gave the a-linked disaccharide $\mathbf{4 0}$ $(0.126 \mathrm{~g}, 13 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{23}+124\left(c 0.93, \mathrm{CHCl}_{3}\right)$ (Found: C, 69.9; H, 5.1\%); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.13\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.6\right.$, $\left.6^{\prime}-\mathrm{H}_{3}\right), 4.42-4.56\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{and} 5^{\prime}-\mathrm{H}\right), 4.66\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{a}} 2.7\right.$, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.3,6-\mathrm{H}^{\mathrm{a}}\right), 4.91\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.3,4-\mathrm{H}\right), 4.93(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{5,6 \mathrm{~b}} 0.5,6-\mathrm{H}^{\mathrm{b}}\right)$, $5.67-5.94\left(6 \mathrm{H}, 1^{\prime}-, 2-, 2^{\prime}-, 3-, 3^{\prime}-\right.$ and $\left.4^{\prime}-\mathrm{H}\right)$, $6.55\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 2.0,1-\mathrm{H}\right)$ and $7.05-8.28(35 \mathrm{H}, \mathrm{m}, 7 \times \mathrm{Ph})$.

## 2,3,4-Tri- $O$-benzoyl- $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -benzoyl- $\alpha$-d-mannopyranose 38

This compound was prepared from compound $37(0.307 \mathrm{~g})$ as described for the hemiacetal derivative 34. FCC [toluene $\longrightarrow$ solvent $C$ ] gave the disaccharide a-hemiacetal 38 ( $0.25 \mathrm{~g}, 90 \%$ ) as an amorphous solid, $[a]_{\mathrm{D}}^{25}+123.3\left(c 1.06, \mathrm{CHCl}_{3}\right)$ (Found: C, 68.1; H, 5.0. $\mathrm{C}_{54} \mathrm{H}_{46} \mathrm{O}_{16}$ requires C, $\left.68.2 ; \mathrm{H}, 4.9 \%\right)$; $\delta_{\mathrm{H}}(200$ $\mathrm{MHz}) 0.83\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.2,6^{\prime}-\mathrm{H}_{3}\right), 3.53\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.07$ $\left(1 \mathrm{H}, \mathrm{d}, J_{1, \mathrm{OH}} 4.1,1-\mathrm{OH}\right), 4.33-4.46\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}^{\mathrm{a}}\right)$, $4.58\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 4.77\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 0.6, J_{6 \mathrm{a}, 6 \mathrm{~b}}\right.$ $\left.12.5,6-\mathrm{H}^{\mathrm{b}}\right), 4.95\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.9,1^{\prime}-\mathrm{H}\right), 5.35\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.8\right.$, $1-\mathrm{H}), 5.42\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.2,3^{\prime}-\mathrm{H}\right), 5.46\left(1 \mathrm{H}, \mathrm{d}, J_{3^{\prime}, 4^{\prime}} 3.2,4^{\prime}-\right.$ H), $5.62-5.75\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 2^{\prime}-\mathrm{H}\right), 5.95\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.0,3-\mathrm{H}\right)$ and $6.98-8.22(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph})$.

## 2,3,4-Tri- $O$-benzoyl- $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -benzoyl- $\alpha$-D-mannopyranosyl hydrogenphosphonate, triethylammonium salt 39

This compound was prepared from the hemiacetal $38(0.208 \mathrm{~g}$, 0.219 mmol ) as described for the H -phosphonate derivative 13. This produced the disaccharide hydrogenphosphonate 39 $(0.232 \mathrm{~g}, 95 \%)$ as a chromatographically homogeneous amorphous solid, $[a]_{\mathrm{D}}^{25}+102.4\left(c 0.96, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ $0.81\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.3,6^{\prime}-\mathrm{H}_{3}\right), 1.27\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 3.01$ $\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeC} H_{2}\right), 3.49\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.37\left(1 \mathrm{H}\right.$, ddd, $J_{5,6 \mathrm{a}}$ $2.7,5-\mathrm{H}), 4.43\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.9,6-\mathrm{H}^{\mathrm{a}}\right), 4.53(1 \mathrm{H}$, $\left.J_{3,4}=J_{4,5}=9.5,4-\mathrm{H}\right), 4.65\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.4,6-\mathrm{H}^{\mathrm{b}}\right), 4.87(1 \mathrm{H}, \mathrm{d}$, $\left.J_{1^{\prime}, 2^{\prime}} 7.9,1^{\prime}-\mathrm{H}\right), 5.34\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.4,3^{\prime}-\mathrm{H}\right), 5.42(1 \mathrm{H}, \mathrm{d}$, $\left.4^{\prime}-\mathrm{H}\right), 5.64\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.4,2^{\prime}-\mathrm{H}\right), 5.67\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 2.0,2-\right.$ $\mathrm{H}), 5.70\left(1 \mathrm{H}, \mathrm{dd}, J_{1, \mathrm{P}} 7.7,1-\mathrm{H}\right), 5.88\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.3,3-\mathrm{H}\right)$, $7.01\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{H}, \mathrm{P}} 636.9, \mathrm{HP}\right)$ and $7.05-8.15(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph})$; $\delta_{\mathrm{P}} 0.11 ; \operatorname{ESMS}(-): m / z 1012.8\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$ (expected $m / z$, 1013.17. $\mathrm{C}_{60} \mathrm{H}_{62} \mathrm{NO}_{18} \mathrm{P}$ requires $M, 1115.37$ ).

## Dec-9-enyl 2,3,4-tri- $O$-benzoyl- $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-d-mannopyranosyl phosphate, triethylammonium salt 41

This compound was prepared by condensation of the H-phosphonate $39(0.12 \mathrm{~g}, 0.107 \mathrm{mmol})$ and dec-9-en-1-ol ( $0.038 \mathrm{~cm}^{3}$, 0.215 mmol ) in pyridine ( $1 \mathrm{~cm}^{3}$ ) in the presence of trimethylacetyl chloride ( $0.033 \mathrm{~cm}^{3}, 0.268 \mathrm{mmol}$ ) followed by oxidation with iodine ( $0.055 \mathrm{~g}, 0.215 \mathrm{mmol}$ ) in pyridine-water ( $95: 5 ; 2$ $\mathrm{cm}^{3}$ ) as described for the synthesis of the phosphodiester $\mathbf{1 5}$. $\mathrm{FCC}\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N},(99: 0: 1) \longrightarrow(89: 10: 1)\right]$ gave the phosphodiester $41(0.108 \mathrm{~g}, 80 \%)$ as an amorphous solid, $[a]_{\mathrm{D}}^{25}$ $+82.2\left(c 1, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 0.84\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.3,6^{\prime}-\mathrm{H}_{3}\right)$, $1.24\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.30\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.58(2 \mathrm{H}, \mathrm{tt}$, $J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2}$ ), 1.97 ( $\left.2 \mathrm{H}, \mathrm{dt}, J 6.9, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 3.05(6 \mathrm{H}$, $\left.\mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.47\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 3.90\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right)$, 4.37-4.49 $\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}^{\mathrm{a}}\right), 4.55\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.4\right.$, $4-\mathrm{H}), 4.64\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.1, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.7,6-\mathrm{H}^{\mathrm{b}}\right), 4.87\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}}\right.$ $\left.7.8,1^{\prime}-\mathrm{H}\right), 4.89\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.3,{ }^{3} J_{\mathrm{H}, \mathrm{H}-\mathrm{Z}} 10.4, \mathrm{HCH}=\mathrm{CH}\right)$, $4.95\left(1 \mathrm{H}, \mathrm{dd},{ }^{3} J_{\mathrm{H}, \mathrm{H}-E} 17.2, \mathrm{HCH}=\mathrm{CH}\right), 5.33\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}} 10.4\right.$, $\left.3^{\prime}-\mathrm{H}\right), 5.43\left(1 \mathrm{H}, \mathrm{d}, J_{3^{\prime}, 4^{\prime}} 3.3,4^{\prime}-\mathrm{H}\right), 5.65\left(1 \mathrm{H}, \mathrm{dd}, J_{1, \mathrm{p}} 8.3,1-\mathrm{H}\right)$, $5.66\left(1 \mathrm{H}, \mathrm{dd}, 2^{\prime}-\mathrm{H}\right), 5.74\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 2.5,2-\mathrm{H}\right), 5.78(1 \mathrm{H}, \mathrm{ddt}$,
$\left.J_{\mathrm{H}, \mathrm{CH}} 6.9, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 5.91\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.4,3-\mathrm{H}\right)$ and $7.05-$ $8.10(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph}) ; \delta_{\mathrm{P}}-2.83$; $\operatorname{ESMS}(-): \mathrm{m} / \mathrm{z} 1166.9$ $\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)\left(\operatorname{expected} m / z, 1167.31 . \mathrm{C}_{70} \mathrm{H}_{80} \mathrm{NO}_{19} \mathrm{P}\right.$ requires $M, 1269.51$ ).

## Dec-9-enyl $\beta$-d-fucopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-d-mannopyranosyl phosphate, triethylammonium salt 9

De- $O$-benzoylation of compound $41(101 \mathrm{mg})$ with 0.05 mol $\mathrm{dm}^{-3} \mathrm{NaOMe}$ in methanol ( 16 h at rt ) followed by work-up, as described in the preparation of the phosphodiester 7, gave the phosphodiester 9 ( $51 \mathrm{mg}, 99 \%$ ) as an amorphous solid, $[a]_{\mathrm{D}}^{28}$ $+22.5(c 0.99, \mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(\mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) 1.22-1.39 $(22 \mathrm{H}, \mathrm{m}$, $6^{\prime}-\mathrm{H}_{3}, 3 \times \mathrm{MeCH}_{2}$ and $\left.5 \times \mathrm{CH}_{2}\right), 1.58\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2}\right.$ $\mathrm{CH}_{2} \mathrm{CH}_{2}$ ), $2.01\left(2 \mathrm{H}, \mathrm{dt}, J 6.9, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 3.15(6 \mathrm{H}$, $\mathrm{q}, 3 \times \mathrm{MeCH}_{2}$ ), $4.38\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.6,1^{\prime}-\mathrm{H}\right), 5.37(1 \mathrm{H}$, br d, $\left.J_{1, \mathrm{P}} 7.1,1-\mathrm{H}\right)$ and $5.83\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{C} H=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}, \delta_{\mathrm{P}}$ and ESMS(-) data: see Table 1.

## 2,3,4-Tri- $O$-benzoyl- $\alpha$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -benzoyl- $\alpha$-D-mannopyranose 42

This compound was prepared from compound $\mathbf{4 0}(160 \mathrm{mg})$ as described for the hemiacetal derivative 34. FCC [toluene $\longrightarrow$ solvent $C$ ] gave the disaccharide a-hemiacetal $\mathbf{4 0}$ ( $99 \mathrm{mg}, 69 \%$ ) as an amorphous solid, $[a]_{\mathrm{D}}^{26}+54.7$ (c 1.02, $\mathrm{CHCl}_{3}$ ) (Found: C, $68.0 ; \mathrm{H}, 5.0 . \mathrm{C}_{54} \mathrm{H}_{46} \mathrm{O}_{16}$ requires C, $\left.68.2 ; \mathrm{H}, 4.9 \%\right) ; \delta_{\mathrm{H}}(200$ $\mathrm{MHz}) 1.13\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.4,6^{\prime}-\mathrm{H}_{3}\right), 3.83\left(1 \mathrm{H}, \mathrm{d}, J_{1, \text { ОН }} 4.3\right.$, $1-\mathrm{OH}), 4.52\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.60\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{~b}} 0.7,5-\mathrm{H}\right), 4.65$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{a}} 2.6,6-\mathrm{H}^{\mathrm{a}}\right), 4.81\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 5.01$ $\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.2,6-\mathrm{H}^{\mathrm{b}}\right), 5.40\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.8,1-\mathrm{H}\right), 5.68(1 \mathrm{H}$, dd, 2-H), $5.75\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.4,3-\mathrm{H}\right), 5.76-5.85\left(4 \mathrm{H}, \mathrm{m}, 1^{\prime}-, 2^{\prime}-\right.$, $3^{\prime}-$ and $\left.4^{\prime}-\mathrm{H}\right)$ and $7.10-8.25(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph})$.

## 2,3,4-Tri- $O$-benzoyl- $\alpha$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$ -benzoyl- $\alpha$-D-mannopyranosyl hydrogenphosphonate, triethylammonium salt 43

This compound was prepared from the hemiacetal $42(88 \mathrm{mg}$, 0.092 mmol ) as described for the H -phosphonate derivative $\mathbf{1 3}$. This produced the disaccharide hydrogenphosphonate 43 (100 $\mathrm{mg}, 97 \%$ ) as a chromatographically homogeneous amorphous solid, $[a]_{\mathrm{D}}^{21}+61\left(c 1.06, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.08\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6}\right.$ $\left.6.3,6^{\prime}-\mathrm{H}_{3}\right), 1.29\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 3.01(6 \mathrm{H}, \mathrm{q}$, $\left.3 \times \mathrm{MeCH}_{2}\right), 4.47\left(1 \mathrm{H}, \mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.57-4.68(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ and $\left.6-\mathrm{H}^{\mathrm{a}}\right), 4.79\left(1 \mathrm{H}, J_{3,4}=J_{4,5}=9.4,4-\mathrm{H}\right), 4.93\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 2.8\right.$, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.9,6-\mathrm{H}^{\mathrm{b}}\right), 5.67\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 2.0, J_{2,3} 3.0,2-\mathrm{H}\right), 5.70-5.82$ $\left(6 \mathrm{H}, \mathrm{m}, 1-, 1^{\prime}-, 2^{\prime}-, 3-, 3^{\prime}-\right.$ and $\left.4^{\prime}-\mathrm{H}\right), 7.10\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{H}, \mathrm{P}} 640.4\right.$, HP) and $6.92-8.23(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph}) ; \delta_{\mathrm{P}} 0.57 ; \operatorname{ESMS}(-): \mathrm{m} / \mathrm{z}$ $1013.0\left(100 \%\right.$, $\left.\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)$(expected $\mathrm{m} / \mathrm{z}$, 1013.17. $\mathrm{C}_{60} \mathrm{H}_{62} \mathrm{NO}_{18} \mathrm{P}$ requires $M, 1115.37$ ).

## Dec-9-enyl 2,3,4-tri- $O$-benzoyl- $\alpha$-d-fucopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-d-mannopyranosyl phosphate, triethylammonium salt 44

This compound was prepared by condensation of the H-phosphonate $43(100 \mathrm{mg}, 0.09 \mathrm{mmol})$ and dec-9-en-1-ol $\left(0.035 \mathrm{~cm}^{3}\right.$, $0.197 \mathrm{mmol})$ in pyridine ( $1 \mathrm{~cm}^{3}$ ) in the presence of trimethylacetyl chloride $\left(0.03 \mathrm{~cm}^{3}, 0.246 \mathrm{mmol}\right)$ followed by oxidation with iodine ( $50 \mathrm{mg}, 0.197 \mathrm{mmol}$ ) in pyridine-water ( $95: 5 ; 2$ $\mathrm{cm}^{3}$ ) as described for the synthesis of the phosphodiester 15. $\mathrm{FCC}\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N},(99: 0: 1) \longrightarrow(87: 12: 1)\right]$ gave the phosphodiester 44 ( $95 \mathrm{mg}, 78 \%$ ) as an amorphous solid, $[a]_{D}^{26}$ $+46\left(c 0.99, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 1.04\left(3 \mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6}, 6.3\right.$, $\left.6^{\prime}-\mathrm{H}_{3}\right), 1.23\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.31\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 1.64$
$\left(2 \mathrm{H}, \mathrm{tt}, J 6.9, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 1.99\left(2 \mathrm{H}, \mathrm{dt}, J 6.9, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right)$, $3.10\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 4.00\left(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 4.44(1 \mathrm{H}$, $\left.\mathrm{q}, 5^{\prime}-\mathrm{H}\right), 4.57-4.70\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}\right.$ and $\left.6-\mathrm{H}^{\mathrm{a}}\right)$, $4.79(1 \mathrm{H}$, $\left.J_{3,4}=J_{4,5}=9.8,4-\mathrm{H}\right), 4.89\left(2 \mathrm{H}\right.$, br d, $6-\mathrm{H}^{\mathrm{b}}$ and $\left.H \mathrm{CH}=\mathrm{CH}\right), 4.96$ $\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.9,{ }^{3} J_{\mathrm{H}, \mathrm{H}-E} 17.0, \mathrm{HCH}=\mathrm{CH}\right), 5.65-5.83(8 \mathrm{H}$, $\mathrm{m}, 1-, 1^{\prime}-, 2-, 2^{\prime}-, 3-, 3^{\prime}-, 4^{\prime}-\mathrm{H}$ and $\left.\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$ and $7.00-$ $8.25(30 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph}) ; \delta_{\mathrm{p}}-2.60$; ESMS(-): m/z 1166.9 $\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}\right]^{-}\right)\left(\operatorname{expected} m / z, 1167.31 . \mathrm{C}_{70} \mathrm{H}_{80} \mathrm{NO}_{19} \mathrm{P}\right.$ requires $M, 1269.51$ ).

## Dec-9-enyl $\alpha$-d-fucopyranosyl-(1 $\rightarrow 4$ )- $\alpha$-D-mannopyranosyl phosphate, triethylammonium salt 10

De- $O$-benzoylation of compound $44(95 \mathrm{mg})$ with 0.05 mol $\mathrm{dm}^{-3} \mathrm{NaOMe}$ in methanol ( 16 h at rt ) followed by work-up, as described in the preparation of the phosphodiester 7, gave the phosphodiester $10(48 \mathrm{mg}, 100 \%)$ as an amorphous solid, $[a]_{D}^{28}$ $+69.8(c 0.98, \mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) 1.23 ( 3 $\left.\mathrm{H}, \mathrm{d}, J_{5^{\prime}, 6^{\prime}} 6.6,6^{\prime}-\mathrm{H}_{3}\right), 1.26-1.40\left(19 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{MeCH}_{2}\right.$ and $\left.5 \times \mathrm{CH}_{2}\right), 1.61\left(2 \mathrm{H}, \mathrm{tt}, J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.04(2 \mathrm{H}, \mathrm{dt}$, $J 7.0, \mathrm{CH}_{2} \mathrm{CH} \mathrm{CH}_{2}=$ ), $3.20\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 5.20(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\left.1^{\prime}-\mathrm{H}\right), 5.42\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J_{1, \mathrm{p}} 7.6,1-\mathrm{H}\right)$ and $5.83(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}, \delta_{\mathrm{P}}$ and $\mathrm{ESMS}(-)$ data: see Table 1.

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[^0]:    $\ddagger$ For methyl $\beta$-D-altropyranoside, $\delta_{\mathrm{C}-5}=75.60 .{ }^{17}$
    $\ddagger$ For methyl $\beta$-D-rhamnopyranoside, $\delta_{\mathrm{C}-5}=73.60 .{ }^{17}$

